

### P100 – Final Exam (May 5, 2009)

Please read the problems carefully and answer them in the space provided. Write on the back of the page, if necessary. Show your work where requested in order to be considered for partial credit. In problems where you are requested to show your work, no credit will be given unless your work is shown.

#### Problem 1 (true or false, each part is worth 2 points):

- a) F 'Speed' and 'velocity' mean the same thing scientifically.
- b) T Protons are made up of other, more fundamental, particles.
- c) F The majority of the heavy elements on Earth were formed during the early stages of the big bang.
- d) F Nuclear fission is the process that produces the energy in stars.
- e) T In quantum mechanics, it is not possible to calculate the exact time that a particular radioactive nucleus will decay, even if you know the half-life for that nucleus.
- f) F The rate of expansion of the universe is currently decreasing.
- g) F A light-year is a unit of time used by astronomers and NASA.
- h) T A Newton is a unit of force.
- i) F Leptons are made up of combinations of two or three quarks.
- j) T According to quantum field theory, the electromagnetic force is conveyed by the exchange of a virtual photon.
- k) T Two nuclei with the same number of protons and different numbers of neutrons are known as 'isotopes'.
- l) F Beta radiation is the same thing as gamma radiation, only slightly less energetic.
- m) T According to modern cosmologists, most of the protons that make up the hydrogen in our universe were formed one microsecond after the big bang started.
- n) F The cosmic microwave background is light that comes to us from the time that the light nuclei were formed during big bang nucleosynthesis.
- o) T The apparent 'flatness' of space is one of the reasons cosmologists advocate theories that incorporate inflation.

#### Problem 2 (3 points):

When you are in a high-flying jet plane, does either your weight or your mass differ from their normal value as measured on the ground?

- a) No, both your weight and your mass are exactly the same as when you are on the ground.
- b) Both your weight and your mass are slightly increased.
- c) Your weight is slightly increased, but your mass is unchanged.
- d) Both your weight and your mass are slightly reduced.
- e) Your mass is slightly reduced, but your weight is unchanged.

#### Problem 3 (3 points):

The acceleration of the moon is

- a) zero, because the moon is obviously not accelerating.
- b)** directed toward Earth.
- c) directed along the moon's forward direction of motion.
- d) directed outward away from Earth.
- e) none of the above.

MISTAKE -  
No correct  
Answer  
All/any answer  
or no answer  
gets full credit

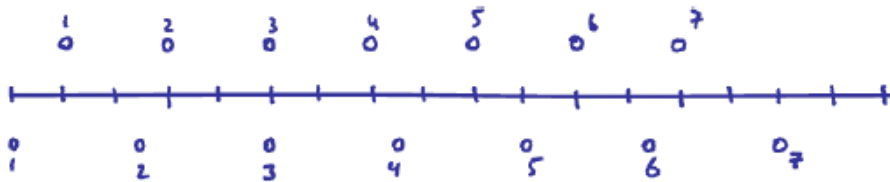
Weight is slightly  
reduced + Mass  
is unchanged

**Problem 4 (3 points):**

According to astrophysicists, Earth originated

- a) as one of Jupiter's moons.
- b) during the big bang, when matter was compressed into stars and planets.
- c) from gases gathered by the sun after the sun had already formed.
- d) when a passing star tore off parts of the sun, and these parts condensed into the planets (including the Earth).
- e) from the same collapsing gas cloud that coalesced into the sun.**

**Problem 5 (3 points):**



The figure represents a multiple-flash photo (with a constant time interval between flashes) of two balls moving to the right, and shows both balls at several numbered times. In the figure,

- ~~a) both balls are accelerated, and the upper ball is moving fastest.~~
- ~~b) neither ball is accelerated, and the upper ball is moving fastest.~~
- ~~c) both balls are accelerated, and the lower ball is moving fastest.~~
- ~~d) the lower ball is accelerated, but the upper ball is not accelerated.~~
- 3 - e) neither ball is accelerated, and the lower ball is moving fastest.**

**Problem 6 (3 points):**



A certain type of atom has only four energy levels, as shown in the figure. The "spectral lines" produced by the element are all visible, except for one ultraviolet (UV) line. The quantum jump that produces the UV line is (*hint: UV light is at a higher frequency than visible light*)

- ~~a) state 2 to 1.~~
- ~~b) state 4 to 1.~~
- ~~c) state 4 to 3.~~
- 3 - d) state 1 to 4.**
- ~~e) impossible to determine without further information.~~

Scores	
1.	30/30
2.	3/3
3.	3/3
4.	3/3
5.	3/3
6.	3/3
7.	3/3
8.	3/3
9.	5/5
10.	5/5
11.	6/6
12.	5/5
13.	5/5
14.	5/5
15.	6/6
16.	6/6
17.	6/6
Total 100/100 😊	

Holy crap!  
Incredible  
Work!

**Problem 7 (3 points):**

Which of the following provides evidence for the quantization of light?

- a) the individual dots seen on a photographic plate when a photo is taken at extremely short exposure times.
- b) interference effects seen when light passes through a narrow opening
- c) both of the above.
- d) experiments that spread light out into a spectrum of colors.
- e) all of the above.

**Problem 8 (3 points):**

According to the Copenhagen view of quantum theory

- a) nature is "non-local," i.e., instantaneously connected across a distance.
- b) individual microscopic events are inherently unpredictable.
- c) both of the above
- d) the overall statistics of large numbers of microscopic events are inherently unpredictable.
- e) all of the above.

**Problem 9 (5 points, show your work):**

You are hanging out with friends on the quad one clear night and see a spectacular meteor (a.k.a a falling star) enter the atmosphere and burn up. With your handy-dandy, special wristwatch, you determine that the meteor was visible to you for exactly 2 seconds. In other words, the time it took for the meteor to enter the atmosphere and burn up was 2 seconds from your point of view. If you were an ant in a fireproof spacesuit riding on the meteor, how long would it have taken for the meteor to enter the atmosphere and burn up from your perspective? Assume the meteor travels at a constant speed of  $0.98c$  during the event. Ignore the effect of gravity on time.

Ant on meteor  
 $t = ?$   
 ← Proper frame - clock moves with meteor  
 time dilation  $\Rightarrow$   
 Time is shortest in the Proper frame

Person on ground  
 $t = 2 \text{ seconds}$   
 at rest on ground

$$\gamma = \frac{1}{\sqrt{1 - \left(\frac{v}{c}\right)^2}}$$

$$\gamma = \frac{1}{\sqrt{1 - (0.98)^2}} \approx 5$$

$\gamma t' = t$  ← you on quad  
 ↑  
 Proper (meteor) frame

$\gamma t' = t$   
 $5 t' = 2 \text{ s}$   
 $t' = \frac{2}{5} \text{ s} = 0.4 \text{ s}$   
 Time as measured by Ant

**Problem 10 (5 points, show your work):**

According to the engineers, a particular airplane must be moving at 30 m/s in order to take off. Suppose this airplane starts from rest and has a constant acceleration of  $5 \text{ m/s}^2$ , how much time does it take for the plane to roll down the runway and achieve takeoff speed?

$$a = \frac{\Delta v}{\Delta t} \quad 5 \frac{\text{m}}{\text{s}^2} = \frac{30 \frac{\text{m}}{\text{s}} - 0 \frac{\text{m}}{\text{s}}}{\Delta t} \quad \Delta t = \frac{30}{5} = 6 \text{ seconds}$$

**Problem 11 (6 points, show your work):**

The energy of a photon is 13 eV. If you were to look at this light as a wave rather than a particle, what is the wavelength of the light corresponding to this photon energy?

$$E = h\nu \rightarrow \nu = \frac{E}{h} \rightarrow \lambda = \frac{c h}{E} = \frac{(3 \times 10^8 \text{ m/s})(4.5 \times 10^{-15} \text{ eV}\cdot\text{s})}{13 \text{ eV}}$$
$$c = \lambda\nu \rightarrow \lambda = \frac{c}{\nu}$$
$$\lambda = 1 \times 10^{-7} \text{ Meters}$$

**Problem 12 (5 pts):**

Briefly explain – using concepts we have discussed in this course - why it is that if you watch a baseball game from a seat in the outfield part of the stadium you see the batter hit the ball around a second before you hear the crack of the bat.

Both sound (what you hear) and light (what you see) are wave phenomenon with finite speeds of propagation. Light moves at roughly  $3 \times 10^8 \text{ m/s}$  while sound moves at  $\sim 330 \text{ m/s}$ . So, from your perspective at a baseball game, the light that allows you to see the batter strike the ball arrives almost instantly while the sound takes almost a second to travel to your ears ... perhaps it is closer to  $\frac{1}{2}$  or  $\frac{1}{3}$  of a second in actuality depending on the location of your seat.

**Problem 13 (5 pts, no need to show work):**

A beam of blue light and a beam of red light are shown on a certain metal plate. A scientist determines that the blue light causes electrons to be emitted by the metal while no electrons are emitted when the red light is shown on the plate. Briefly explain why this happens using concepts we have discussed in this course.

Blue light has a higher frequency than red light. Since the photon energy is given by  $E = h\nu$ , the photon energy for blue light is greater than that for red light. If the metal on which the light is shown requires more than the photon energy of red light but less than the photon energy of blue light to eject the electron, it would explain the observations.

**Problem 14 (5 pts):**

Please explain, using concepts from this course, why it is that in collisions between SUV's and small cars, the death rate for the occupants in the small cars is much higher than for the occupants of the SUVs. (In formulating your answer, please ignore the effect of the differences in height of the vehicles even though it has a very real effect on the death rates. I'm looking for you to discuss another aspect of the collisions that leads to the death rate difference.)

During the collision each object exerts the same force on the other according to Newton's third law (For every force, there is an equal and opposite force). Since  $F = ma \Rightarrow M_{SUV} a_{SUV} = M_{small car} a_{small car}$

The mass of the SUV is much larger than the mass of the small car. So

**Problem 15 (6 pts):** the acceleration of the SUV (and the person inside the SUV) is

Carbon-14 ( $^{14}C$ ) undergoes radioactive beta decay (emitting an electron) with a half-life of roughly 5700 years. What nuclear species (isotope) results from this decay?

Much less, leading to less damage to persons inside the vehicle.



Atomic number (# of protons) increases by 1



Problem 16 (6 pts):

Clearly Answers to 16, 17 can vary.

Briefly defend the statement: "Science is very different from art."

Science is a methodology used to understand the world around us. Art is not so constrained. Sometimes art is a product of the artist's attempts to understand the world and other times the motivation might come from another source such as a desire to share beauty or entertain. To the extent art is used to probe the human understanding of the universe it is not constrained by experiment as is science. Also art can be perceived differently by different people while every attempt is made to insure scientific observations/claims are seen/agreed to by everyone.

Problem 17 (6 pts):

Briefly defend the statement: "Science is very similar to art."

Often art is used as a way for humans to understand and share insights about the world around us, as is science. Also, science is a human endeavor. As such, it is subject to human bias <sup>and</sup> human interpretation, as is art. Science may not be as open to interpretation as art, but it is hardly as black and white as it is typically portrayed. Scientists are drawn to formulate hypotheses, theories and interpretations that they perceive as aesthetically pleasing when possible. A scientist learns to judge the science of others in much the same way an artist might appreciate an artistic work. Often scientists will describe scientific work using words you might consider more appropriately <sup>used</sup> by an artist... "elegant" or "crisp" or "beautiful" work.

hydrogen 1 <b>H</b> 1.0079	beryllium 4 <b>Be</b> 9.0122	helium 2 <b>He</b> 4.0026	lithium 3 <b>Li</b> 6.941	boron 5 <b>B</b> 10.811	carbon 6 <b>C</b> 12.011	nitrogen 7 <b>N</b> 14.007	oxygen 8 <b>O</b> 15.999	fluorine 9 <b>F</b> 18.998	neon 10 <b>Ne</b> 20.180
sodium 11 <b>Na</b> 22.990	magnesium 12 <b>Mg</b> 24.305	argon 18 <b>Ar</b> 39.948	potassium 19 <b>K</b> 39.098	aluminum 13 <b>Al</b> 26.982	silicon 14 <b>Si</b> 28.086	phosphorus 15 <b>P</b> 30.974	sulfur 16 <b>S</b> 32.065	chlorine 17 <b>Cl</b> 35.453	krpton 36 <b>Kr</b> 83.80
calcium 20 <b>Ca</b> 40.078	scandium 21 <b>Sc</b> 44.956	zinc 30 <b>Zn</b> 65.39	strontium 38 <b>Sr</b> 87.62	gallium 31 <b>Ga</b> 69.723	germanium 32 <b>Ge</b> 72.61	arsenic 33 <b>As</b> 74.922	selenium 34 <b>Se</b> 78.96	bromine 35 <b>Br</b> 79.904	xenon 54 <b>Xe</b> 131.29
barium 56 <b>Ba</b> 137.33	yttrium 39 <b>Y</b> 88.906	copper 29 <b>Cu</b> 63.546	radium 88 <b>Ra</b> [226]	nickel 28 <b>Ni</b> 58.693	tin 50 <b>Sn</b> 118.71	antimony 51 <b>Sb</b> 121.76	tellurium 52 <b>Te</b> 127.60	iodine 53 <b>I</b> 126.90	radon 86 <b>Rn</b> [222]
cesium 55 <b>Cs</b> 132.91	zirconium 40 <b>Zr</b> 91.224	silver 47 <b>Ag</b> 107.87	actinide series	cadmium 48 <b>Cd</b> 112.41	lead 82 <b>Pb</b> 207.2	bismuth 83 <b>Bi</b> 208.98	polonium 84 <b>Po</b> [209]	astatine 85 <b>At</b> [210]	
francium 87 <b>Fr</b> [223]	niobium 41 <b>Nb</b> 92.906	gold 79 <b>Au</b> 196.97	lanthanide series	mercury 80 <b>Hg</b> 200.59	ununquadium 114 <b>Uuq</b> [289]	thallium 81 <b>Tl</b> 204.38	atlanium [209]		
	hafnium 72 <b>Hf</b> 178.49	platinum 78 <b>Pt</b> 195.08		unumbium 112 <b>Uub</b> [277]					
	tantalum 73 <b>Ta</b> 180.95	iridium 77 <b>Ir</b> 192.22		ununium 111 <b>Uun</b> [271]					
	tungsten 74 <b>W</b> 183.84	rhodium 45 <b>Rh</b> 102.91		ununium 110 <b>Uun</b> [271]					
	seaborgium 106 <b>Sg</b> [266]	rhodium 45 <b>Rh</b> 102.91		ununium 109 <b>Uun</b> [268]					
	dubnium 105 <b>Db</b> [262]	osmium 76 <b>Os</b> 190.23		ununium 108 <b>Uun</b> [269]					
	bohrium 107 <b>Bh</b> [264]	iridium 77 <b>Ir</b> 192.22		ununium 107 <b>Uun</b> [268]					
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	hassium 108 <b>Hs</b> [269]	gold 79 <b>Au</b> 196.97		ununium 105 <b>Uun</b> [268]					
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				ununium -16 <b>Uun</b> [148]					

Some potentially useful formulas

$$F = \frac{G m_1 m_2}{r^2} \left[ \begin{array}{l} m_1 \text{ and } m_2 \text{ in kg} \\ r \text{ in meters} \\ F \text{ in Newtons} \end{array} \right] \rightarrow G = 6.7 \times 10^{-11}$$

$$F = \frac{k q_1 q_2}{r^2} \left[ \begin{array}{l} q_1, q_2 \text{ in Coulombs} \\ r \text{ in meters} \\ F \text{ in Newtons} \end{array} \right] \rightarrow k = 9 \times 10^9$$

$$F = ma$$

$$(\text{distance}) = (\text{Speed})(\text{time})$$

$$v = \frac{\Delta x}{\Delta t}$$

$$a = \frac{\Delta v}{\Delta t}$$

$$\text{Work} = \text{force} \times \text{distance}$$

$$\text{Momentum} = p = mv$$

$$\Delta x' = \gamma \Delta x, \Delta x \text{ longest in proper frame}$$

$$\Delta t' = \gamma \Delta t, \Delta t \text{ shortest in proper frame}$$

$$\gamma = \frac{1}{\sqrt{1 - (v/c)^2}}$$

$$1 \text{ Joule} = 1.6 \times 10^{-19} \text{ eV}$$

$$\text{speed of Sound} = 330 \text{ m/s}$$

$$c = 3 \times 10^8 \text{ m/s}$$

$$h = 6.6 \times 10^{-34} \text{ J}\cdot\text{s}$$

$$\text{or } 4.5 \times 10^{-15} \text{ eV}\cdot\text{s}$$

$$\lambda = \frac{h}{p} = \frac{h}{mv}$$

$$v = \lambda \nu$$

$$\nu = \frac{1}{T} \quad (T = \text{period})$$

$$E = h\nu$$

$$\Delta x \Delta p \geq \frac{h}{2\pi} \quad \Delta E \Delta t \geq \frac{h}{2\pi}$$

$$\frac{\Delta N}{\Delta t} = \lambda N \quad t_{1/2} = 0.693/\lambda$$