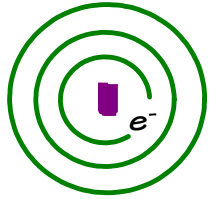


Physics 123, April 3, 2013

①

- Exam 2 looms large Apr. 11, Dewey 1101, 0800
- Up to start of Bohr's model of atom (Apr. 1 lecture)
- With all the crazy optics stuff...
Can use Both Sides of an 8.5 x 11 inch sheet
for notes + Formulas
I will provide a formula sheet as well
- More on Material coverage in email soon ...
But it will cover from end of Exam 1 material
to start of Bohr Model



Bohr model

quantized circular orbits

$$r_n = \frac{n^2 \hbar^2}{kze^2 m}$$

$$n = 1, 2, 3, \dots$$

with quantized energies

$$E_{n \text{ Total}} = - \frac{mk^2 z^2 e^4}{2n^2 \hbar^2}$$

- e^- in circular orbits
- e^- held in atom via Coulomb attraction
- Single e^-
- e^- exists in discrete stable orbits
- Photons emitted as e^- jumps between orbits

- Photon energy corresponds to difference in energy between the two energy levels

②

er ... last class I forgot to mention something important ^③

Bohr's model works!!

For single e^- atoms Bohr's model reproduces fairly well the experimentally determined discrete atomic spectra!

NOT perfect ... revisions made

e.g., allow nucleus to move + have e^- orbit center-of-mass

elliptical orbits

⋮

- Bohr model gets people to take Rutherford's nuclear atom seriously
- relates quanta of light and quantization of Atomic Energy levels
- Generally a decent starting place for intuitions

But

- Fails to describe spectra of larger atoms
- Treats e^- classically (can think of de Broglie motivating the quantization ... But this mixes wave + particle for e^-)
- Fails to describe detailed structure ("fine" structure, Zeeman Splitting ... etc.)

Successful model of atom must come from a theory that treats e^- as wave from the ground up \rightarrow quantum mechanics

Particles & waves leads to a strange new world

(5)

Recall from waves + Fourier analysis

(A) Can describe waves in position or spatial frequency
 x $k = 2\pi/\lambda$

If wave is narrow in x , require higher frequency components

(B) Can describe waves in time or temporal frequency
 t $\omega = 2\pi/T$

If wave is narrow in t , require higher frequency components

So, for all waves $\Delta x \Delta k \sim 1$ and $\Delta t \Delta \omega \sim 1$
width in position space Δx width in frequency space Δk

⑥

$$\text{De Broglie} \rightsquigarrow p = \frac{h}{\lambda} = h \frac{k}{2\pi} = \hbar k \rightsquigarrow \Delta k = \frac{\Delta p}{\hbar}$$

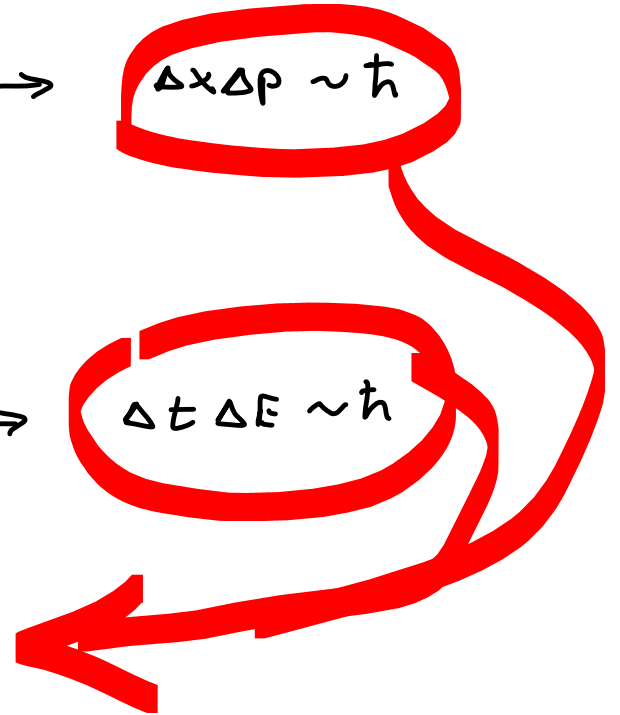
$\hbar \equiv \frac{h}{2\pi}$

$$\Delta x \Delta k \sim 1 \rightarrow \Delta x \frac{\Delta p}{\hbar} \sim 1 \rightarrow \Delta x \Delta p \sim \hbar$$

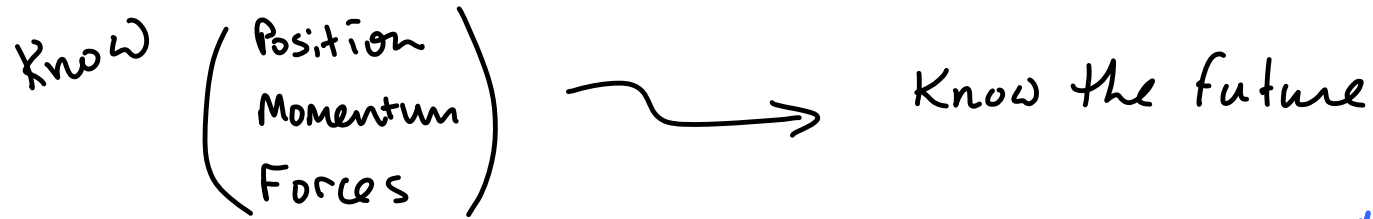
$$E = h\nu = \hbar\omega \rightarrow \Delta\omega = \frac{\Delta E}{\hbar}$$

$$\Delta t \Delta\omega \sim 1 \rightarrow \Delta t \frac{\Delta E}{\hbar} \sim 1 \rightarrow \Delta t \Delta E \sim \hbar$$

Two forms of
Heisenberg's Uncertainty Principle



In world of Newton and Particles 'R Particles:



Universe is deterministic!

Not so in world of Particles 'R Waves

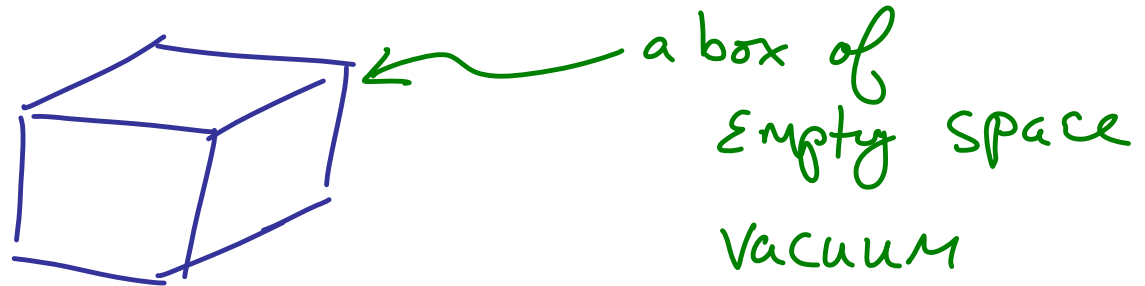
The better I determine the position
The less well I know the momentum
+ vice versa

Inherent Uncertainty!



Mourn the loss of Deterministic universe!
Rejoice in the new freedom!

The uncertainty Principle
opens up Harry Potterish possibilities



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consider this bit of empty space over a very short time
Energy over that time is very uncertain

Can have fluctuations in the energy \rightarrow Particle-Antiparticle
Pairs

Virtual particles, fleeting existence

The vacuum is a seething mess of virtual particles



Quantum Mechanics

(10)

recall $\vec{F} = -\vec{\nabla} V$

Need a Wave equation that allows us to calculate the motion of a particle under the influence of a force (Moving in a "potential")

Why not use $\nabla^2 \psi = \frac{1}{c^2} \frac{\partial^2 \psi}{\partial t^2}$?

This is for free traveling wave (no force)

Also need consistency w/ $E = h\nu$
 $P = h/\lambda$

want:

■ consistency w/ $E = h\nu$, $P = h/\lambda$

■ Energy conservation including potential

$$E = \frac{P^2}{2m} + V$$

■ Linear in $\psi(x,t)$ in order to satisfy Superposition

$$\begin{array}{l}
 \textcircled{A} \quad E = h\nu = \hbar\omega \\
 \quad \quad p = \frac{h}{\lambda} = \hbar k
 \end{array}
 \left. \vphantom{\begin{array}{l} E = h\nu = \hbar\omega \\ p = \frac{h}{\lambda} = \hbar k \end{array}} \right\} \rightarrow \textcircled{B} \quad E = \frac{p^2}{2m} + V \rightarrow \textcircled{C} \quad \hbar\omega = \underbrace{\frac{\hbar^2 k^2}{2m}} + V \quad \textcircled{II}$$

Let's assume particle is a 1-d wave, Amplitude = 1
 $i(kx - \omega t)$

$$\textcircled{D} \quad \psi(x, t) = e^{i(kx - \omega t)}$$

$$\left. \begin{array}{l} \textcircled{C} \\ \textcircled{D} \end{array} \right\} \rightarrow \textcircled{E} \quad -\frac{\hbar}{i} \frac{\partial \psi}{\partial t} = -\frac{\hbar^2}{2m} \frac{\partial^2 \psi}{\partial x^2} + V\psi$$

← mult by i/i

1-d nonrelativistic
 time-dependent
 Schrödinger
 equation

$$i\hbar \frac{\partial \psi(x, t)}{\partial t} = -\frac{\hbar^2}{2m} \frac{\partial^2 \psi(x, t)}{\partial x^2} + V\psi(x, t)$$

Two of the
Big players
in the early
development of
QM

Werner Karl Heisenberg
(1901 - 1976)

Nobel Prize in physics - 1932
for "the creation of quantum
Mechanics"

(Max Born, Pascual Jordan - co-workers)



(12)

Matrix
formulation



Erwin Rudolf Josef Alexander Schrödinger
(1887 - 1961) Austria

1933 Nobel Prize in physics

1926 - Paper on wave Mechanics of Matter
Annalen der Physik

"for discovery of new and productive forms of
atomic theory"

Wave equation
formulation

$$-i\hbar \frac{\partial \Psi(x,t)}{\partial t} = -\frac{\hbar^2}{2m} \frac{\partial^2 \Psi(x,t)}{\partial x^2} + V(x,t) \Psi(x,t)$$

What do you do with this?
 Put in the physical situation $V(x,t)$
 Solve for $\Psi(x,t)$ and Energy

(13)

Will do examples shortly

but let me carry on for a bit with more background/theory

Can imagine it is useful to consider Static Situations

$$\text{let } V(x,t) \longrightarrow V(x)$$

Potential is constant in time

$$\Psi(x,t) \longrightarrow \Psi(x) \Psi(t)$$

Assume space + time dependence are separable

$$-i\hbar \frac{\partial \Psi(x) \Psi(t)}{\partial t} = -\frac{\hbar^2}{2m} \frac{\partial^2 \Psi(x) \Psi(t)}{\partial x^2} + V(x) \Psi(x) \Psi(t)$$

$$-i\hbar \Psi(x) \frac{\partial \Psi(t)}{\partial t} = -\frac{\hbar^2}{2m} \Psi(t) \frac{\partial^2 \Psi(x)}{\partial x^2} + V(x) \Psi(x) \Psi(t)$$

÷ by $\psi(x)\psi(t)$

$$\underbrace{-i\hbar \frac{1}{\psi(t)} \frac{\partial \psi(t)}{\partial t}}_{\text{Function of } t \text{ only}} = \underbrace{\frac{-\hbar^2}{2m} \frac{1}{\psi(x)} \frac{\partial^2 \psi(x)}{\partial x^2} + V(x)}_{\text{Function of } x \text{ only}} \quad (14)$$

Equality only holds for all x and t

if both sides are equal to same constant
 \leadsto choose it to be E

(A)

$$\underbrace{-i\hbar \frac{1}{\psi(t)} \frac{\partial \psi(t)}{\partial t}}_{\text{Function of } t \text{ only}} = E$$

one soln of this is $\psi(t) = e^{-i\frac{E}{\hbar}t}$

And (B)

$$\underbrace{\frac{-\hbar^2}{2m} \frac{1}{\psi(x)} \frac{\partial^2 \psi(x)}{\partial x^2} + V(x)}_{\text{Function of } x \text{ only}} = E$$

for an 1d wave $\Psi(x,t) = e^{i(kx - \omega t)} = \Psi(x)\Psi(t) = e^{ikx} e^{-i\omega t}$

recall $E = \hbar\omega$

$$\Psi(x,t) = \Psi(x) e^{-i\frac{E}{\hbar}t}$$

$$-\frac{\hbar^2}{2m} \frac{1}{\Psi(x)} \frac{\partial^2 \Psi(x)}{\partial x^2} + V(x) = E$$

Time independent
1d nonrelativistic
Schrödinger eqn

$$-\frac{\hbar^2}{2m} \frac{\partial^2 \Psi(x)}{\partial x^2} + V(x)\Psi(x) = E\Psi(x)$$

Woohoo! We got wave equations ...

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But what's waving?

What is $\Psi(x,t)$? it is complex (i.e., has $i = \sqrt{-1}$ in it
as opposed to being complicated)
So Association of Ψ w/
real world is NOT obvious

Max Born - 1926 - Born interpretation



Square of Ψ represents
the probability density
for the particle/wave

Born's postulate : at time t , $\Psi^*(x,t) \Psi(x,t) dx = |\Psi(x,t)|^2 dx$
gives the probability of finding the particle
between x and $x+dx$

(17)

Normalize $\Psi(x,t)$ so that the total probability is 1

$$\int_{-\infty}^{\infty} |\Psi(x,t)|^2 dx = 1$$

The particle must be
SomeWhere

in 3d ... $\int |\Psi(\vec{r},t)|^2 dV = 1$...

Are we having fun yet? Need to do a couple of examples (18)

Example: 1d Free particle

$$V(x) = 0 \text{ for all } x$$

Time ind.
Schr. eqn

$$-\frac{\hbar^2}{2m} \frac{\partial^2 \psi(x)}{\partial x^2} + V(x) \psi(x) = E \psi(x) \quad (A)$$

Try $\psi(x, t) = e^{i(kx - \frac{E}{\hbar} t)} = e^{ikx} e^{-i\frac{E}{\hbar} t}$

as a solution

$$\psi(x) = e^{ikx}$$

$$\frac{\hbar^2 k^2}{2m} e^{ikx} = E e^{ikx}$$

$$\text{works if } k = \sqrt{\frac{2mE}{\hbar^2}}$$

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$\psi(x,t)$ here is a wave traveling in $+x$ direction

$$\psi(x) = e^{-ikx} \text{ also a soln } \dots k = \sqrt{\frac{2mE}{\hbar^2}}$$

$\psi(x,t)$ here is a wave traveling in $-x$ direction

So, general solution

$$\psi(x) = Ae^{ikx} + Be^{-ikx}$$

where $k^2 = \frac{2mE}{\hbar^2}$

or equivalently

$$\psi(x) = A' \sin(kx) + B' \cos(kx)$$

$$k^2 = \frac{2mE}{\hbar^2}$$

You will find that it is easiest to solve different types of problems using one form or the other

$$e^{i\theta} = \frac{e^{i\theta} + e^{-i\theta}}{2} + i \frac{e^{i\theta} - e^{-i\theta}}{2}$$

recall Euler's Formula

$$e^{i\theta} = \cos\theta + i\sin\theta$$

$$\frac{e^{i\theta} - e^{-i\theta}}{2}$$

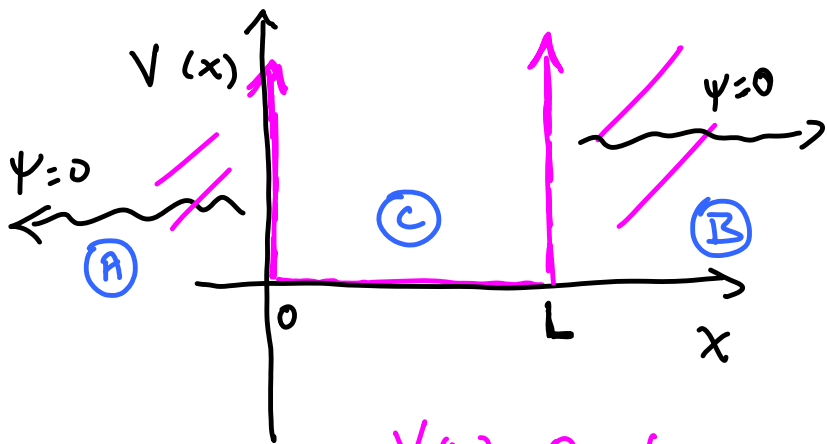
$$A' \sin kx + B' \cos kx$$

$$\frac{A'}{2} (e^{ikx} - e^{-ikx}) + \frac{B'}{2} (e^{ikx} + e^{-ikx})$$

$$\underbrace{\left(\frac{A'}{2} + \frac{B'}{2} \right)}_{\equiv A} e^{ikx} + \underbrace{\left(\frac{B'}{2} - \frac{A'}{2} \right)}_{\equiv B} e^{-ikx}$$

1-d Particle in a box

infinite Square Well potential



$$V(x) = 0 \text{ for } 0 < x < L$$

$$V(x) = \infty \text{ for } x < 0, L < x$$

as $V \rightarrow \infty$ Schr eqn makes no sense
 so $\psi = 0$ there

for $0 < x < L$ $V(x) = 0$
 particle is free
 inside the Box

$$\psi(x) = A \sin kx + B \cos kx$$

Use Boundary conditions to constrain Solution:

at $x=0$, $\psi(x)=0$

$$0 = A \sin kx + B \cos kx$$

$\underbrace{\hspace{10em}} \rightarrow B=0$

(22)
 $\psi(x) = A \sin kx$

at $x=L$, $\psi(x)=0$

$$0 = A \sin kL$$

$$kL = n\pi \quad n=1, 2, 3 \dots$$

$n=0$ not considered because for $n=0$ either $k=0$ $\psi(x)=0$
or $L=0$ no box

$$\psi(x) = A \sin\left(\frac{n\pi x}{L}\right)$$

$$n=1, 2, 3 \dots$$

in region $0 < x < L$

$$\textcircled{A} \quad k = \frac{n\pi}{L} \quad \text{So,} \quad \textcircled{B} \quad E_n = \frac{n^2 \pi^2 \hbar^2}{L^2 2m} \quad \text{where } n=1, 2, 3, \dots \quad \textcircled{23}$$

energy is quantized! You'll find this is always true
in attractive potentials
as in wanting to bind or capture

$$E_1 = \frac{\pi^2 \hbar^2}{L^2 2m}$$

$$E_2 = \frac{4\pi^2 \hbar^2}{L^2 2m} \quad \dots$$



→ This is the lowest energy allowed
Note that it is NOT zero as we
would have in classical physics

(24)

Determine A by using normalization to probability of 1

$$\int \psi^*(x) \psi(x) = 1 \quad \rightsquigarrow \quad \int_0^L A^2 \sin^2(kx) dx = 1 \quad \text{(A)}$$

Done
in Giancoli:
ex 38-6
p. 1032

$$A^2 \int_0^L \sin^2(kx) dx = 1$$

← → = $\frac{L}{2}$

$$A = \sqrt{\frac{2}{L}}$$

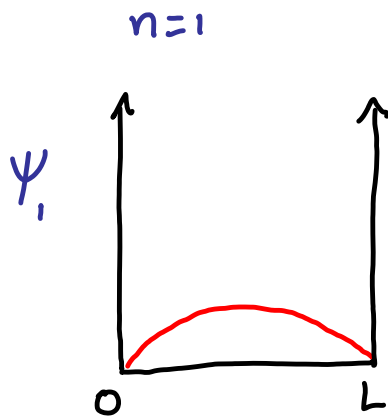
∞ 1d square well
Wave function

$$\psi_n(x) = \sqrt{\frac{2}{L}} \sin\left(\frac{n\pi}{L}x\right) \quad \text{for } n = 1, 2, 3, \dots \quad \text{in } 0 < x < L$$

What does the wavefunction look like?

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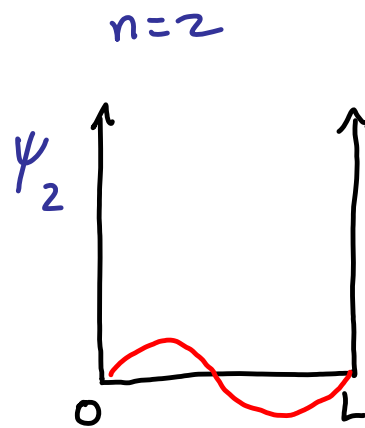
$$kl = n\pi$$
$$2\pi/L = n\pi$$
$$\lambda = 2L/n$$



$$\lambda = 2L$$

$$\psi_1 = \sqrt{\frac{2}{L}} \sin \frac{\pi}{L} x$$

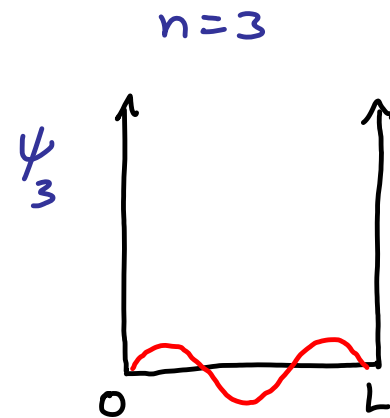
(A)



$$\lambda = L$$

$$\psi_2 = \sqrt{\frac{2}{L}} \sin \frac{2\pi}{L} x$$

(B)



$$\lambda = \frac{2L}{3}$$

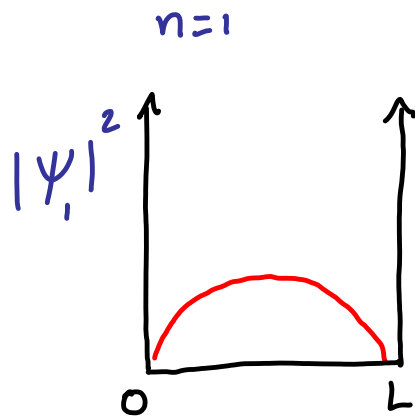
$$\psi_3 = \sqrt{\frac{2}{L}} \sin \frac{3\pi}{L} x$$

(C)

...

...

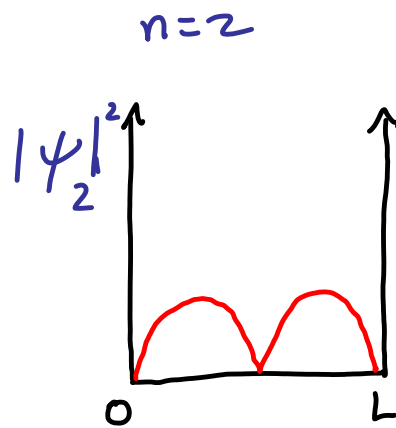
What is the probability density for seeing particle at position x (26)



$$\lambda = 2L$$

$$|\psi_1|^2 = \frac{2}{L} \sin^2 \frac{\pi x}{L}$$

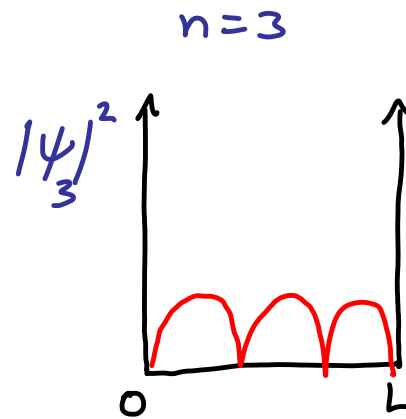
(A)



$$\lambda = L$$

$$|\psi_2|^2 = \frac{2}{L} \sin^2 \frac{2\pi x}{L}$$

(B)



$$\lambda = \frac{2L}{3}$$

$$|\psi_3|^2 = \frac{2}{L} \sin^2 \frac{3\pi x}{L}$$

(C)

...

...