

Physics 123 - April 17, 2013

Today - Quantum Weirdness
Giancoli, chapt 38

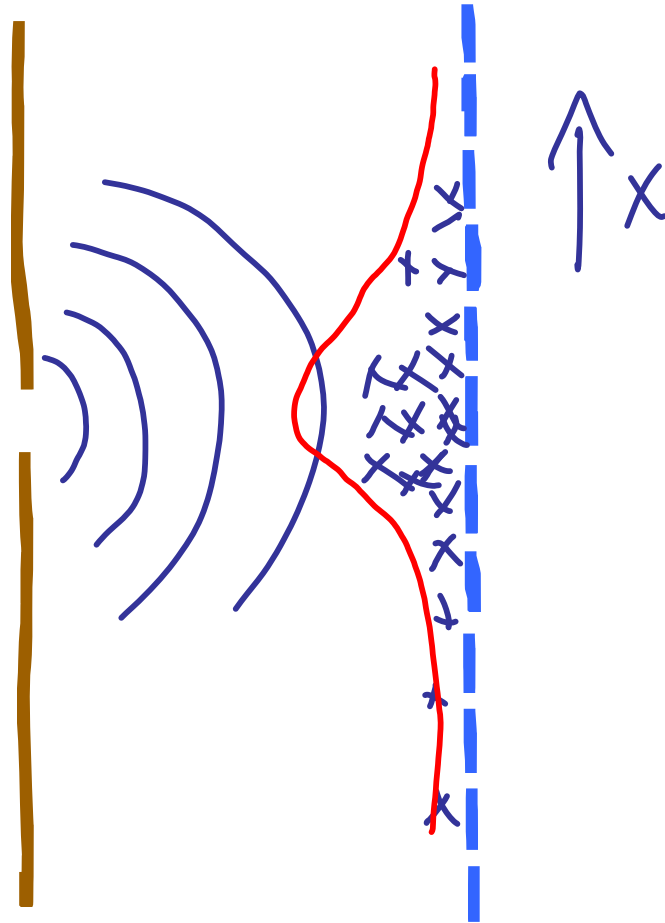
Check out article on BB

Byrne - The Many Worlds
of Hugh

Today + Monday - H atom
Many electron atoms
Giancoli, chapt 39

Quantum
Wierdness

e^- →



← Probability

$\Psi(x)$ NOT well defined

$\Psi^2(x)$ is well defined → probability

Max Born German (1882-1970)

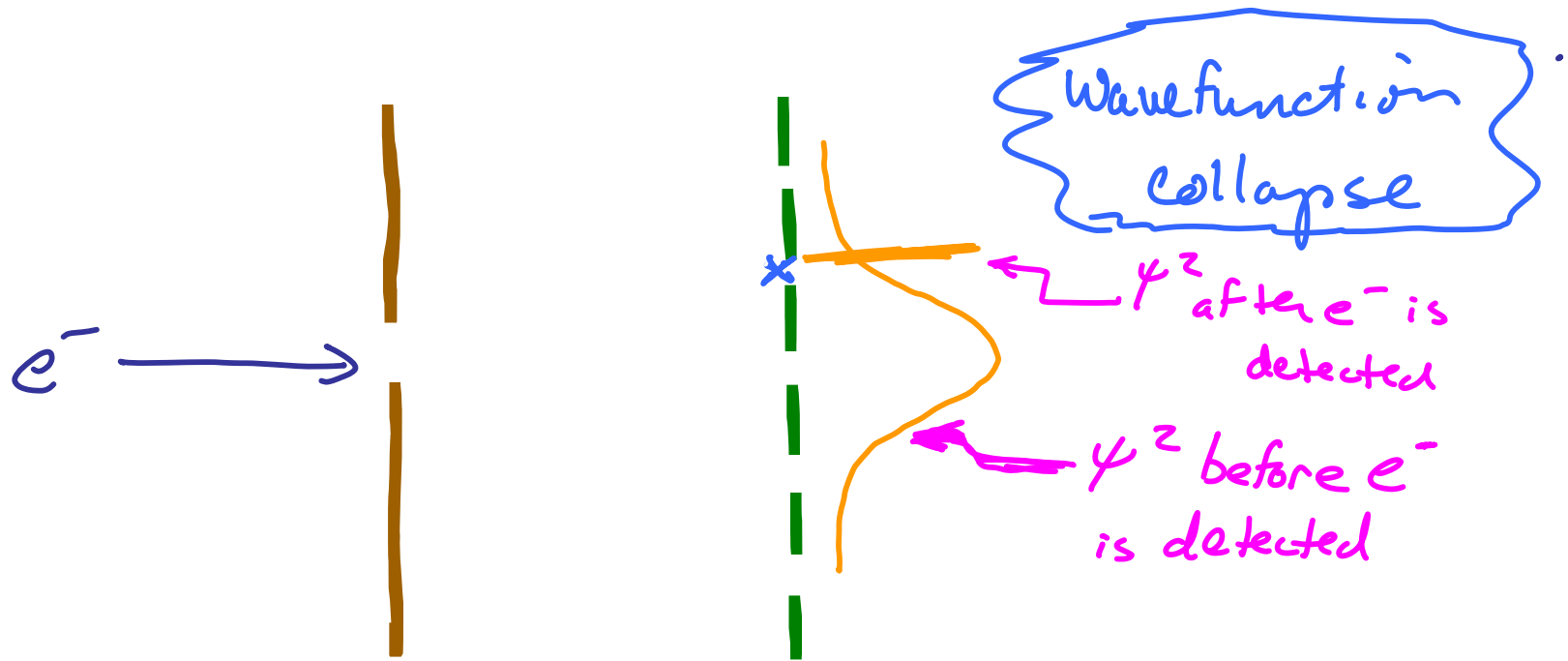


1954 Nobel Prize in physics

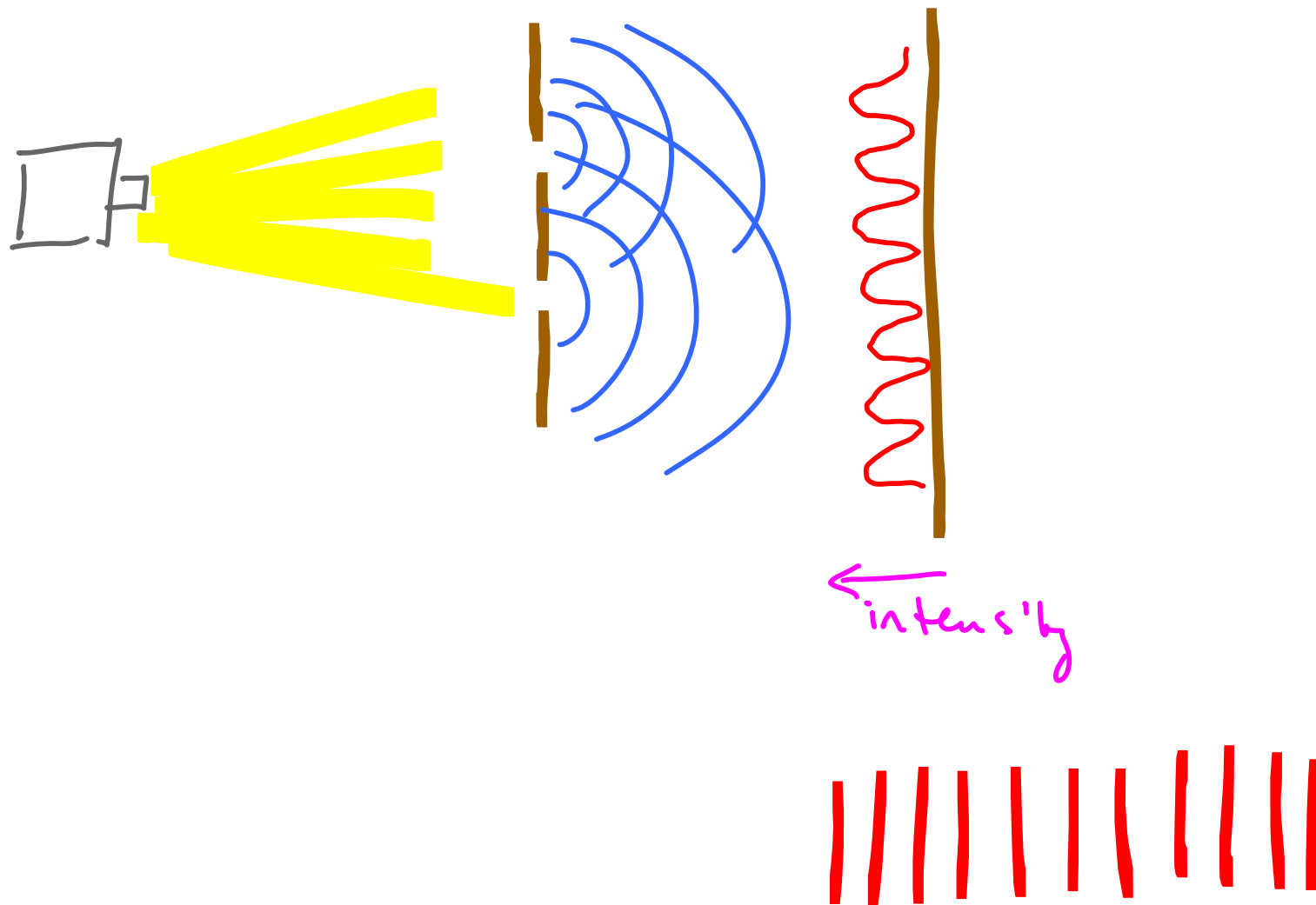
"For his fundamental research
in quantum mechanics,
especially for his statistical
interpretation of the
wavefunction"

$\psi(x)$ wave function

$\psi^2(x) \sim$ probability of finding particle
in region of space

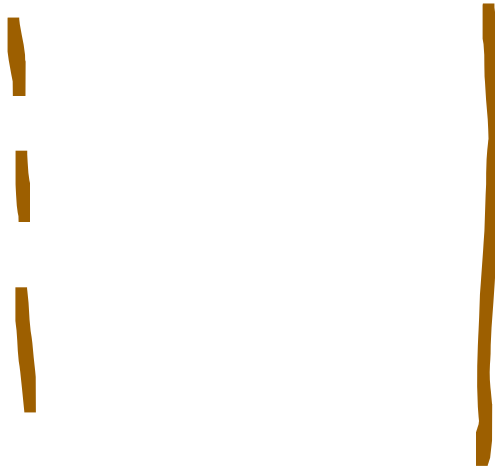


Once electron hits the film/detector we know with 100% certainty where the electron hits
- So wavefunction has to "collapse"



get same even if 1 photon at
a time

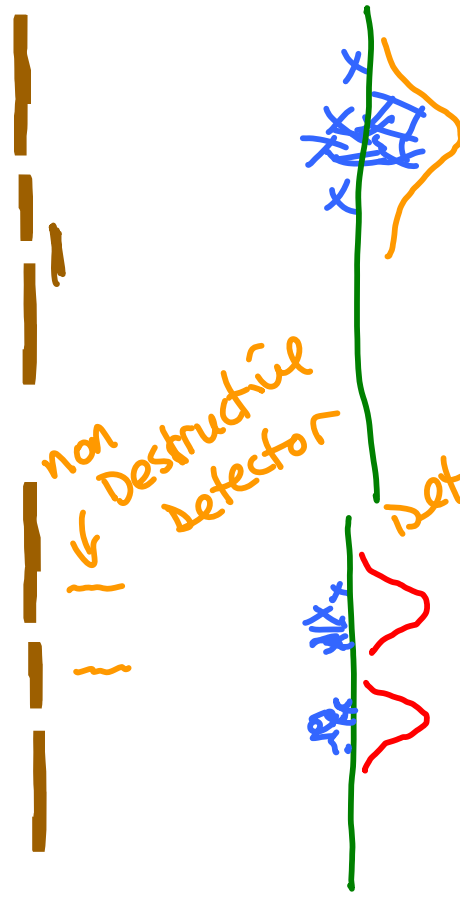
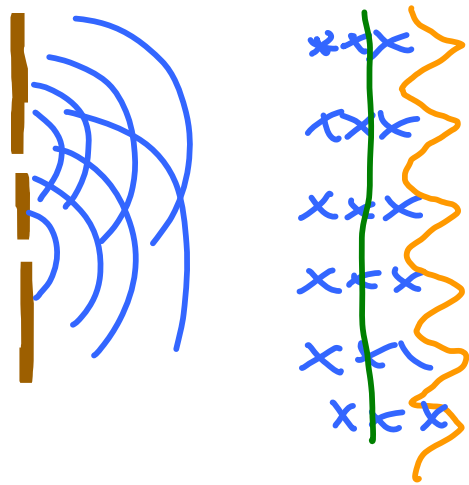
$e^- \rightarrow$



Get Same even if $1e^-$ at a time



just determining the slit that the e^- passes thru (even with non-destructive detector) is sufficient to collapse the wave function.



Happens even if particle already passed slit

non destructive detector

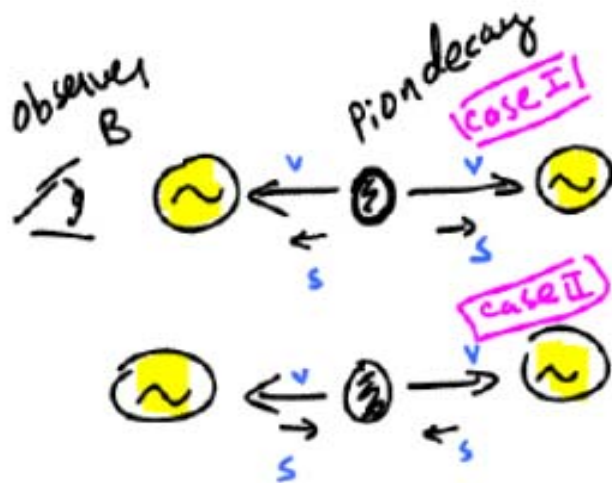
detector on

detector off

EPR Paradox — Einstein, Podolski, Rosen
1935

8

"Spooky Action at a distance"



photon spin = 1



Two photons are produced at once — They are correlated.

If one has spin one way the other has spin the other way.

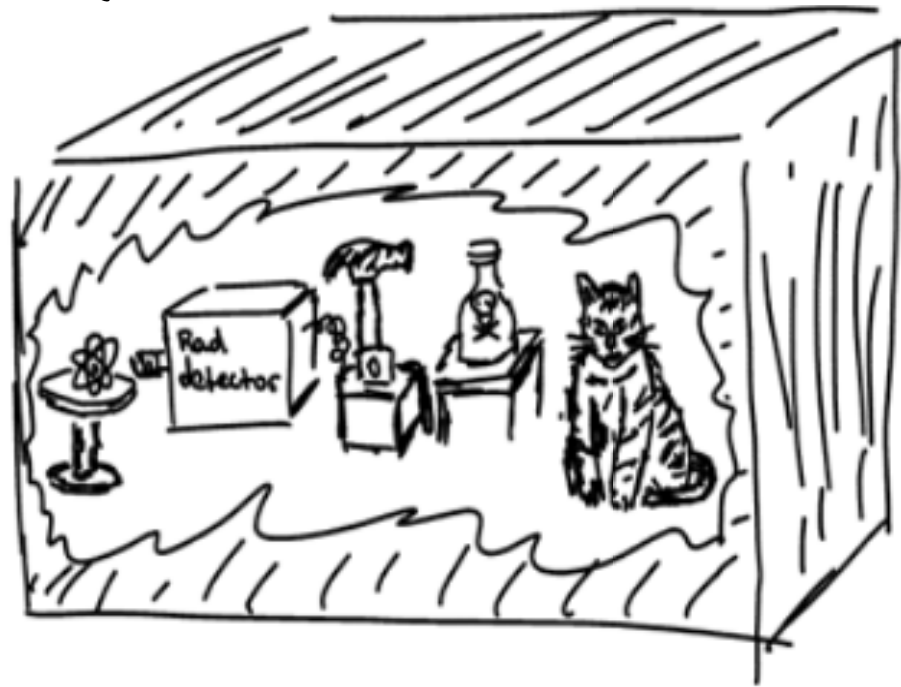
They are in an "entangled quantum STATE"

When observer A observes the spin of photon — The wavefunction collapses and the spin of the photon observer B will observe is determined.

But collapse instantaneous and observers A + B far apart

Does this mean information conveys faster than speed of light?

Schrödinger's Cat



Thought experiment
nucleus has
50:50 chance
of decaying +
killing the cat.
What is the
"state" of the
cat before box
opened?

Copenhagen
Interpretation

$$\text{nucleus quantum state} = \frac{1}{2}(\text{decayed}) + \frac{1}{2}(\text{not decayed})$$



$$\text{cat state} = \frac{1}{2}(\text{dead}) + \frac{1}{2}(\text{alive})$$



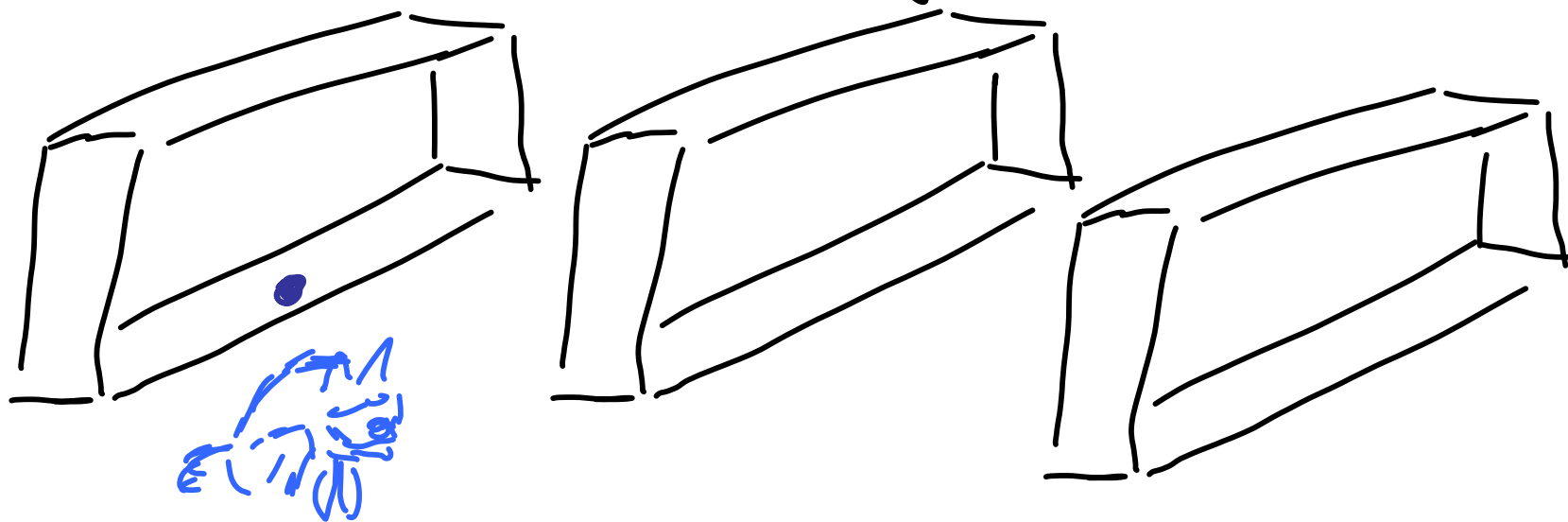


Hugh
Everett (1957)

Bryce DeWitt
1960's + 70's
↳ Many
Worlds
interpretation

Overall wavefunction does not collapse. IT evolves in time.
"Decoherence" forces wavefunction to evolve into different
streams that do NOT interact.

Dog, ball, shelf analogy

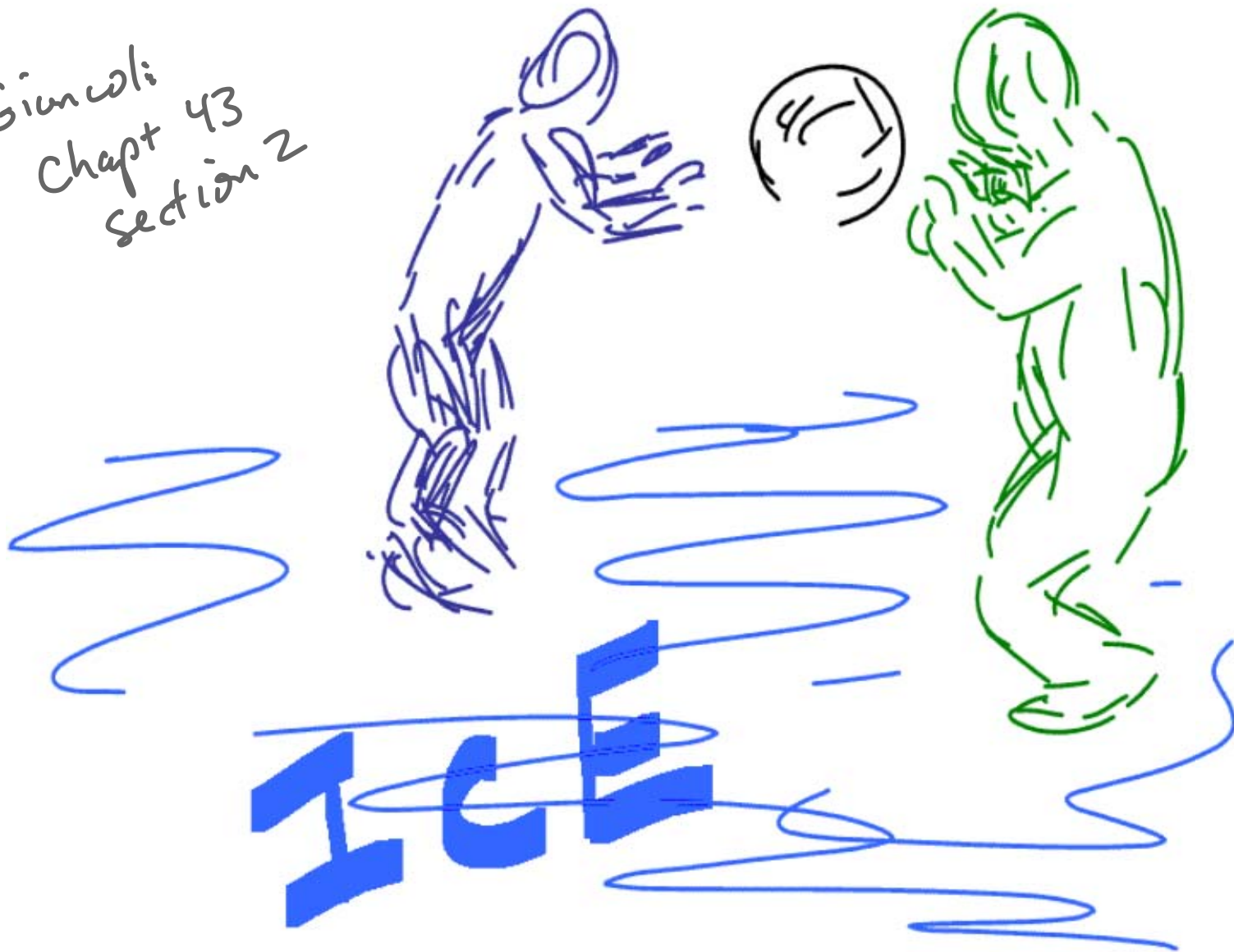


ψ might include both upper and lower shelves

But dog only experiences lower shelves

QUANTUM Field Theory \rightarrow Exchange force

Giunco
Chapt 43
Section 2



Time Independent Schrödinger equation (1d, nonrelativistic)

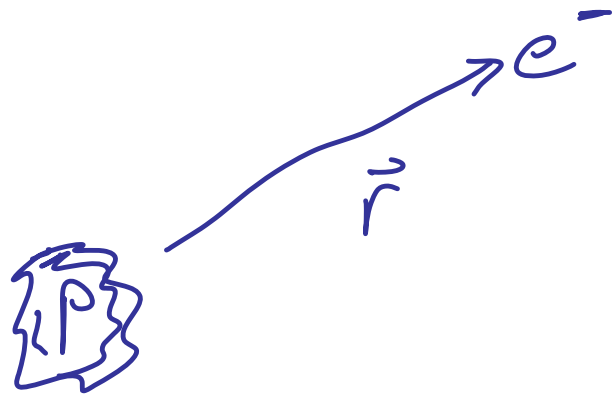
$$-\frac{\hbar^2}{2m} \frac{\partial^2 \Psi(x)}{\partial x^2} + V(x) \Psi(x) = E \Psi(x)$$

$|\Psi(x)|^2 dv \equiv$ probability of finding particle in dv

$$\int_{\substack{\text{all} \\ \text{space}}} |\Psi(x)|^2 dv = 1 \quad \text{Particle is } \underline{\underline{\text{Someplace}}}$$

Sub in for V , solve for Ψ and E

H atom



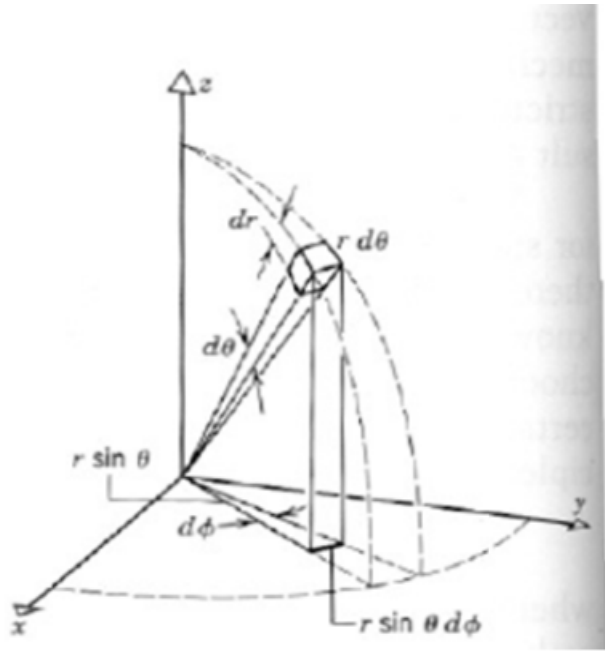
$$V(r) = \frac{1}{4\pi\epsilon_0} \frac{|e|^2}{r^2}$$

Schr. eqn in 3d

$$-\frac{\hbar^2}{2m} \nabla^2 \psi + V\psi = E\psi$$

plug in V and solve

need to use spherical polar coordinates



$$\psi \longrightarrow \psi(r, \theta, \phi)$$

$$V(r) = -\frac{1}{4\pi\epsilon_0} \frac{|e|^2}{r^2}$$

$$-\frac{\hbar^2}{2m} \left[\frac{1}{r^2} \frac{\partial}{\partial r} r^2 \frac{\partial \psi}{\partial r} + \frac{1}{r^2 \sin^2 \theta} \frac{\partial^2 \psi}{\partial \phi^2} + \frac{1}{r^2 \sin \theta} \frac{\partial}{\partial \theta} \sin \theta \frac{\partial \psi}{\partial \theta} \right] - \frac{1}{4\pi\epsilon_0} \frac{|e|^2}{r^2} \psi = E \psi$$

Now solve



Can be solved exactly ...

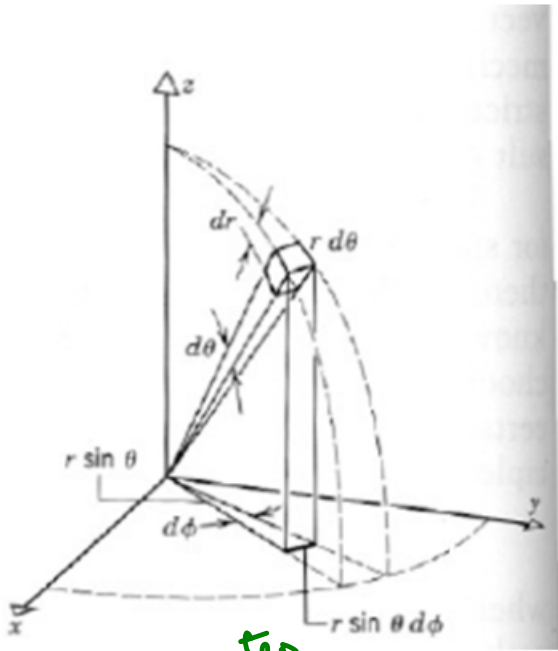
$$\Psi(r, \theta, \phi) \rightarrow \Psi(R) \Psi(\Theta) \Psi(\Phi)$$

Can separate Schr. equation into eqns for $\Psi(R)$, $\Psi(\Theta)$, $\Psi(\Phi)$ + Solve

"Energy" or "principle" quantum # $n = 1, 2, 3 \dots$

"orbital" quantum # $l = 0, 1, \dots, n-1$

"Magnetic" quantum # $-l, -(l-1), \dots, 0, 1, \dots, l-1, l$



Bound system
 \rightarrow Discrete states
 R, Θ, Φ parts tied together
 $n \rightarrow$ Similar to Bohr's n

Table 7.1 Some Hydrogen Atom Wave Functions

| n | l | m_l | $R(r)$ | $\Theta(\theta)$ | $\Phi(\phi)$ |
|-----|-----|---------|---|----------------------------------|---------------------------------------|
| 1 | 0 | 0 | $\frac{2}{a_0^{3/2}} e^{-r/a_0}$ | $\frac{1}{\sqrt{2}}$ | $\frac{1}{\sqrt{2\pi}}$ |
| 2 | 0 | 0 | $\frac{1}{(2a_0)^{3/2}} \left(2 - \frac{r}{a_0}\right) e^{-r/2a_0}$ | $\frac{1}{\sqrt{2}}$ | $\frac{1}{\sqrt{2\pi}}$ |
| 2 | 1 | 0 | $\frac{1}{\sqrt{3}(2a_0)^{3/2}} \frac{r}{a_0} e^{-r/2a_0}$ | $\sqrt{\frac{3}{2}} \cos \theta$ | $\frac{1}{\sqrt{2\pi}}$ |
| 2 | 1 | ± 1 | $\frac{1}{\sqrt{3}(2a_0)^{3/2}} \frac{r}{a_0} e^{-r/2a_0}$ | $\frac{\sqrt{3}}{2} \sin \theta$ | $\frac{1}{\sqrt{2\pi}} e^{\pm i\phi}$ |



P.A.M. Dirac - on the development of quantum mechanics

"The underlying laws necessary for the mathematical theory of a large part of physics and the whole of chemistry are thus completely known."