

## Canonical Ensemble

Consider a system of interest in contact with a thermal reservoir - ie system is separated from reservoir by a fixed impermeable but thermally conducting wall. Let  $E$  be the energy of the system, and  $E_T - E$  be the energy of the reservoir. Total energy  $E_T$  is fixed. Then

$$\Omega_T(E_T) = \int \frac{dE}{\Delta} \underbrace{\Omega(E)}_{\text{system}} \underbrace{\Omega_R(E_T - E)}_{\text{reservoir}}$$

The prob that the system has energy  $E$  is therefore

$$P(E) \propto \Omega(E) \Omega_R(E_T - E) = \Omega(E) e^{S_R(E_T - E)/k_B}$$

since the reservoir is large,  $E \ll E_T$ , so we can expand

$$\begin{aligned} \Omega_R(E_T - E) &\approx \exp \frac{1}{k_B} \left\{ S_R(E_T) - \frac{\partial S_R}{\partial E_R}(E) + \dots \right\} \\ &= \exp \frac{1}{k_B} \left\{ S_R(E_T) - \frac{E}{T} \right\} = \text{const} e^{-E/k_B T} \end{aligned}$$

$$\text{so } P(E) \propto \Omega(E) e^{-E/k_B T} = e^{S(E)/k_B - E/k_B T}$$

$$\Rightarrow \boxed{P(E) = \frac{\Omega(E) e^{-E/k_B T}}{\Delta Q_N(T, V)}} \quad \text{normalization} \quad \int dE P(E) = 1$$

where

$$\boxed{Q_N(T, V) = \int \frac{dE}{\Delta} \Omega(E) e^{-E/k_B T}} \quad \text{is the canonical partition function}$$

if energy levels are discrete  $Q_N(T, V) = \sum_{E_i} \Omega(E_i) e^{-E_i/k_B T}$

Using the density of states

$$g(E) = \frac{1}{N!} \int \frac{dg_i}{h^{3N}} \int dp_i \delta(H(q_i, p_i) - E)$$

↳ Gibbs correction due to indistinguishable particles

$$\Omega(E) = g(E)\Delta$$

then

$$Q_N(T, V) = \int dE g(E) e^{-E/k_B T}$$

$$\text{Probability } P(E) = \frac{g(E) e^{-E/k_B T}}{Q_N(T, V)}$$

Combining the above

$$Q_N(T, V) = \frac{1}{N!} \int \frac{dg_i}{h^{3N}} \int dt_i e^{-H(q_i, p_i)/k_B T}$$

Boltzmann factor  $e^{-H/k_B T}$

The density of states  $g(E)$  has built into it all the information about a system as far as its thermodynamic behavior is concerned.

If one knows  $g(E)$ , then one can compute

$$\text{the entropy } S(E) = k_B \ln \Omega(E) = k_B \ln [g(E)\Delta]$$

Or one can compute the canonical partition function

$$Q_N(T, V) = \int dE g(E) e^{-E/k_B T}$$

In the canonical ensemble, the energy of the system of interest is not fixed, but follows a probability distribution set by a fixed temperature. The temperature determines the average energy. The canonical ensemble does not depend on any details of the reservoir, except its being large!

In thermodynamics we saw that when one wishes to use  $T$  as the variable instead of  $S$ , the potential to use is the Helmholtz free energy  $A(T, V, N)$ . We will now see that there is a direct relation between  $A$  and the canonical partition function  $Q_N$ .

In the canonical ensemble, the average energy is:

$$\langle E \rangle = \int dE E P(E)$$

average value  $\uparrow$ 
 $\uparrow$  probability density to have energy  $E$

$$P(E) = \frac{g(E) e^{-E/k_B T}}{Q_N(T, V)}, \quad Q_N = \int dE g(E) e^{-E/k_B T}$$

define  $\beta \equiv 1/k_B T$  then

$$\langle E \rangle = \frac{\int dE E g(E) e^{-\beta E}}{\int dE g(E) e^{-\beta E}} = -\frac{\partial}{\partial \beta} \left[ \frac{\int dE g(E) e^{-\beta E}}{\int dE g(E) e^{-\beta E}} \right]$$

$$\Rightarrow \langle E \rangle = -\frac{\partial}{\partial \beta} \ln \left[ \int dE g(E) e^{-\beta E} \right] = -\frac{\partial}{\partial \beta} \ln Q_N(T, V)$$

$$\boxed{\langle E \rangle = -\frac{\partial}{\partial \beta} \ln Q_N(T, V)}$$

## Relation between $Q_N$ and Helmholtz free energy $A$

$$A = E - TS \Rightarrow E = A + TS = A - T \left( \frac{\partial A}{\partial T} \right)_{V, N}$$

$$E = A - T \left( \frac{\partial A}{\partial T} \right)_{V, N} = -T^2 \left[ \frac{\partial}{\partial T} \left( \frac{A}{T} \right) \right]_{V, N}$$

$$= \left( \frac{\partial (A/T)}{\partial (1/T)} \right)_{V, N} = \left( \frac{\partial (\beta A)}{\partial \beta} \right)_{V, N}$$

$$\rightarrow E = - \frac{\partial}{\partial \beta} (-\beta A)$$

Compare with  $\langle E \rangle = - \frac{\partial}{\partial \beta} (\ln Q_N)$

$$\Rightarrow \boxed{A(T, V, N) = -k_B T \ln Q_N(T, V)}$$

Or:  $A = E - TS$

$$\Rightarrow -\frac{A}{T} = S - \left(\frac{1}{T}\right) E$$

$\Rightarrow -\frac{A}{T}$  is Legendre transform of  $S$  with respect to  $E$

$$\Rightarrow \frac{\partial (-A/T)}{\partial (1/T)} = -E$$

$$\Rightarrow \frac{\partial (A/T)}{\partial (1/T)} = E$$

Another way to write the above is

$$-\frac{A}{T} = k_B \ln Q_N$$

Note:  $-\frac{A}{T}$  is the Legendre transform of  $S$  with respect to  $E$

~~Given  $S(E, V, N)$ ,  $\left(\frac{\partial S}{\partial E}\right)_{V, N} = \frac{1}{T}$~~

~~$\Rightarrow$  Legendre transform of  $S$  with respect to  $E$  is~~

~~$S - \frac{E}{T} = \frac{1}{T} (TS - E) = -\frac{A}{T}$~~

Now Compare

$$-\frac{A}{T} = k_B \ln Q_N$$

$Q_N$  is canonical partition function

$$S = k_B \ln \Omega$$

$\Omega$  is microcanonical partition function

the thermodynamic potential is the log of the partition function

$$-\frac{A}{T} = S - \frac{E}{T}$$

$-\frac{A}{T}$  is Legendre transform of  $S$

$$Q_N = \int \frac{dE}{\Delta} \Omega(E) e^{-\beta E}$$

$Q_N$  is Laplace transform of  $\Omega$

This holds more generally: If one takes the Laplace transform of a partition function, the corresponding thermodynamic potential of the new partition function is just the Legendre transform of the original thermo potential.

There is still one point to check out more carefully

$$E = -\frac{\partial}{\partial \beta} (-\beta A)$$

is a result within the microcanonical ensemble

i.e. we started with  $S(E, V, N)$  with  $E$  fixed and Legendre transformed to get  $A$

$$\langle E \rangle = -\frac{\partial}{\partial \beta} \ln Q_N$$

is a result within the canonical ensemble, with

$E$  fluctuating - only the average  $E$  is fixed by the temperature  $T$ .

We expect these must be the same since fluctuations about  $\langle E \rangle$  should vanish in thermo limit of very large systems

Alternatively:

$-\frac{A}{T}$  computed from the Legendre transform of  $S$  is the microcanonical Helmholtz free energy

$-\frac{A}{T} \equiv k_B \ln Q_N$  computed from the canonical partition function  $Q_N$  is the canonical Helmholtz free energy

How do we know the two are really the same?

In other words, how do we know that the thermodynamic properties computed within the ~~microcanonical~~ microcanonical ensemble will agree with the thermodynamic properties computed within the canonical ensemble?

How do we know that the two ensembles give equivalent results?

The results will be equivalent if the fluctuations of  $E$  about its average  $\langle E \rangle$  can be ignored.

We will see that this is in fact the case in the "thermodynamic limit" of  $N \rightarrow \infty$ .

## Energy fluctuations

- In canonical ensemble,  $E$  is not fixed, but has a prob distr.  
How wide is the distr in  $E$ ?

Consider

$$\frac{\partial \langle E \rangle}{\partial \beta} = \frac{\partial}{\partial \beta} \left[ \frac{\int \frac{dE}{\Delta} E \Omega(E) e^{-\beta E}}{Q_N} \right]$$

$$= \frac{\int \frac{dE}{\Delta} E \Omega(E) e^{-\beta E} (-E)}{Q_N} - \frac{\int \frac{dE}{\Delta} E \Omega(E) e^{-\beta E}}{Q_N^2} \frac{\partial Q_N}{\partial \beta}$$

$$= -\langle E^2 \rangle - \langle E \rangle \frac{\partial \ln Q_N}{\partial \beta}$$

$$= -\langle E^2 \rangle + \langle E \rangle^2$$

So the fluctuation in the energy  $E$  is:

$$\langle E^2 \rangle - \langle E \rangle^2 = -\frac{\partial \langle E \rangle}{\partial \beta} = \frac{\partial^2}{\partial \beta^2} (\ln Q_N) = -\frac{\partial^2}{\partial \beta^2} (\beta A)$$

Note:  $\langle (E - \langle E \rangle)^2 \rangle = \langle E^2 - 2E\langle E \rangle + \langle E \rangle^2 \rangle$

$$= \langle E^2 \rangle - 2\langle E \rangle \langle E \rangle + \langle E \rangle^2 = \langle E^2 \rangle - \langle E \rangle^2$$

So fluctuation in energy is

$$\langle (E - \langle E \rangle)^2 \rangle = \langle E^2 \rangle - \langle E \rangle^2 = -\frac{\partial \langle E \rangle}{\partial \beta} = -k_B \frac{\partial \langle E \rangle}{\partial (1/T)} = k_B T^2 \frac{\partial \langle E \rangle}{\partial T}$$

$$= k_B T^2 C_V$$

↑ specific heat at constant volume

Note:  $\langle E \rangle \sim N$ ,  $C_V \sim N$

$$\Rightarrow \frac{\sqrt{\langle E^2 \rangle - \langle E \rangle^2}}{\langle E \rangle} = \frac{\sqrt{k_B T^2 C_V}}{\langle E \rangle} \sim \frac{\sqrt{N}}{N} = \frac{1}{\sqrt{N}} \rightarrow 0 \text{ as } N \rightarrow \infty$$