

example: The ideal monatomic gas

$$\begin{aligned}\text{From expt: } PV &= N k_B T \quad \Rightarrow \quad \frac{P}{T} = \frac{N}{V} k_B = \frac{k_B}{v} \\ E &= \frac{3}{2} N k_B T \quad \Rightarrow \quad \frac{1}{T} = \frac{3}{2} k_B \frac{N}{E} = \frac{3}{2} \frac{k_B}{u}\end{aligned}$$

if we can find μ , then we have entropy S via

$$S = \frac{E}{T} + \frac{P}{T} V - \frac{\mu}{T} N$$

From Gibbs-Duhem relation in entropy representation

$$\begin{aligned}d\left(\frac{\mu}{T}\right) &= u d\left(\frac{1}{T}\right) + v d\left(\frac{P}{T}\right) \\ &= u \frac{3}{2} k_B d\left(\frac{1}{u}\right) + v k_B d\left(\frac{1}{v}\right)\end{aligned}$$

$$d\left(\frac{\mu}{T}\right) = -\frac{3}{2} \frac{k_B}{u} du - \frac{k_B}{v} dv$$

integrate to get

$$\left(\frac{\mu}{T}\right) - \left(\frac{\mu}{T}\right)_0 = -\frac{3}{2} k_B \ln\left(\frac{u}{u_0}\right) - k_B \ln\left(\frac{v}{v_0}\right)$$

where u_0 at v_0 are some reference state, and $\left(\frac{\mu}{T}\right)_0$ is an unknown constant of integration. Then one gets

$$\begin{aligned}S &= \frac{E}{T} + \frac{P}{T} V - \frac{\mu}{T} N = \frac{3}{2} \frac{k_B E}{(E/N)} + \frac{k_B V}{(V/N)} \\ &\quad + \frac{3}{2} N k_B \ln\left(\frac{u}{u_0}\right) + k_B N \ln\left(\frac{v}{v_0}\right) - \left(\frac{\mu}{T}\right)_0 N\end{aligned}$$

$$S = \frac{3}{2} k_B N + k_B N - \left(\frac{\mu}{T}\right)_0 N + N k_B \ln\left[\left(\frac{u}{u_0}\right)^{3/2} \left(\frac{v}{v_0}\right)\right]$$

use

$$E = uN, E_0 \equiv u_0 N_0, V = vN, V_0 \equiv v_0 N_0$$

$$\Rightarrow S(E, V, N) = \frac{N}{N_0} S_0 + N k_B \ln \left[\left(\frac{E}{E_0} \right)^{3/2} \left(\frac{V}{V_0} \right) \left(\frac{N}{N_0} \right)^{-5/2} \right]$$

where $S_0 = \frac{5}{2} k_B N_0 - \left(\frac{\mu}{T} \right)_0 N_0$ is a constant

So from experimental knowledge of two of the equations of state $\frac{1}{T}$ and $\frac{P}{T}$ as functions of E, N, V , we have derived the entropy $S(E, V, N)$. All behaviour of the ideal gas can now be deduced from knowledge of S .

Alternatively, we could derive $a = \frac{S}{N}$ as follows:

$$E = TS - PV + \mu N \Rightarrow S = \frac{E}{T} + \frac{P}{T} V - \frac{\mu}{T} N$$

$$\Rightarrow a = \frac{u}{T} + \frac{P}{T} v - \frac{\mu}{T} \quad \text{where } u = \frac{E}{N}, v = \frac{V}{N}$$

$$da = \frac{1}{T} du + \frac{P}{T} dv + \underbrace{u d\left(\frac{1}{T}\right) + v d\left(\frac{P}{T}\right) - d\left(\frac{\mu}{T}\right)}_{\text{these cancel due to the Gibbs-Duhem relation as expressed in the entropy formulation}}$$

So

$$\Rightarrow da = \frac{1}{T} du + \frac{P}{T} dv$$

$$ds = \frac{1}{T} du + \frac{p}{T} dv$$

$$= \frac{3}{2} \frac{k_B}{u} du + \frac{k_B}{v} dv$$

$$\text{since } \begin{cases} \frac{1}{T} = \frac{3}{2} \frac{k_B}{u} \\ \frac{p}{T} = \frac{k_B}{v} \end{cases}$$

integrate

$$s - s_0 = \frac{3}{2} k_B \ln(u/u_0) + k_B \ln(v/v_0)$$

$$s = s_0 + k_B \ln \left[\left(\frac{u}{u_0} \right)^{3/2} \left(\frac{v}{v_0} \right) \right]$$

$$\text{substitute in } S = Ns, \quad E = Nu, \quad V = Nv$$

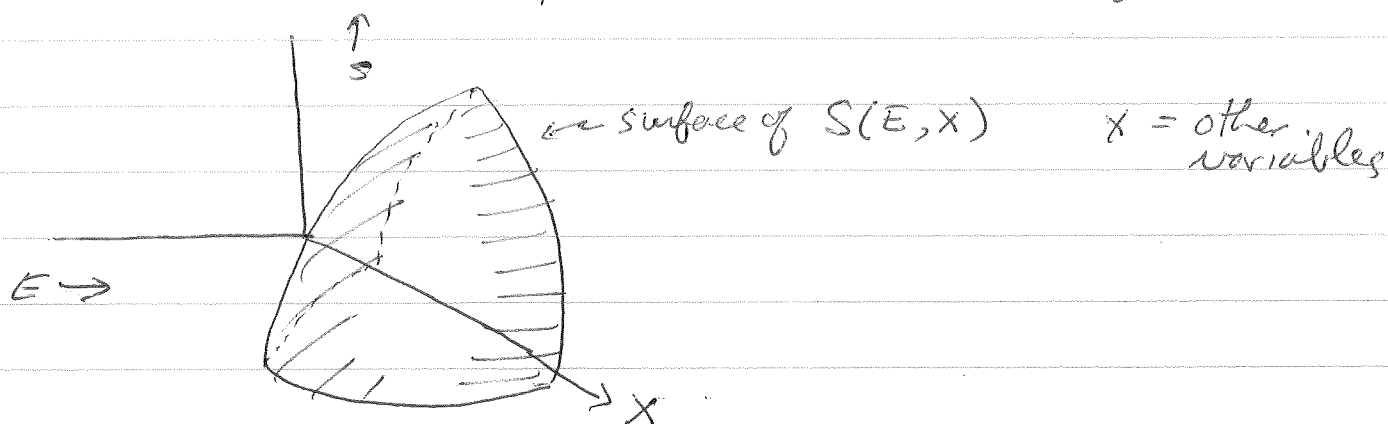
$$S_0 = N_0 s_0, \quad E_0 = N_0 u_0, \quad V_0 = N_0 v_0$$

and we recover the earlier result for $S(E, V, N)$

Energy Minimum Principle

Postulate II stated that when constraints are removed, the equilibrium state will be the one that maximizes the entropy $S(E, V, N, \dots)$

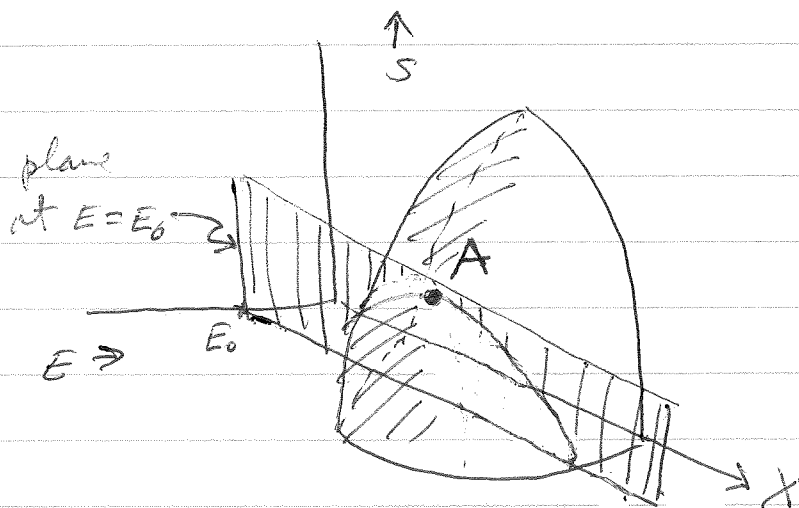
We saw that entropy is a ~~convex~~^{concave} function of its variables



For a situation where the total E is held fixed, at value E_0 , then if X is an unconstrained degree of freedom, it will take in equilibrium that value X_0 that maximizes S for the given fixed E_0 . This is determined by the intersection of the surface $S(E, X)$ with the plane at fixed $E = E_0$.

X_0 is given by the point A that maximizes S along this curve of intersection

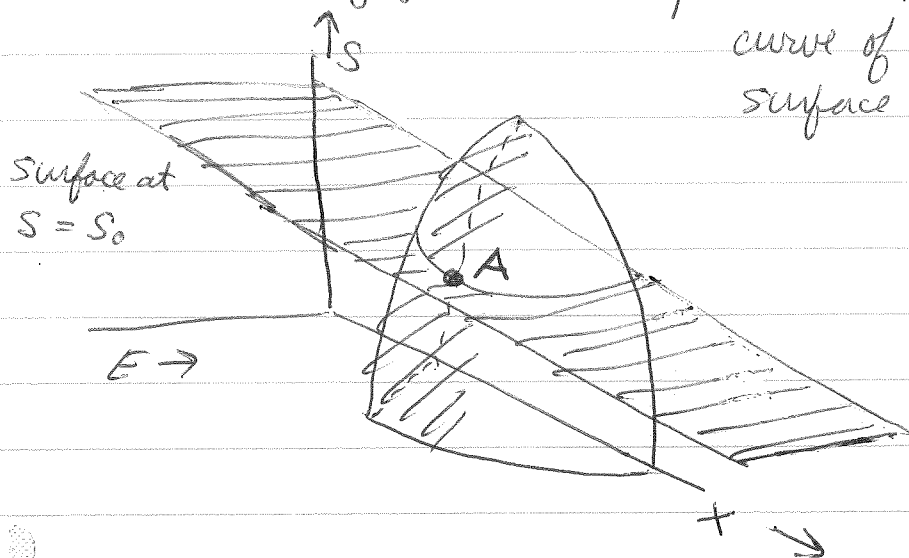
$$S_0 = \max_X [S(E_0, X)]$$



Callen Fig 5.1

Suppose now an alternative situation in which the total entropy S is held fixed at value S_0 .

Then if X is an unconstrained degree of freedom we see that the equilibrium state at S_0, E_0 corresponds to minimizing the energy with respect to X , along the curve of intersection between $S(E, X)$ surface and plane of const $S = S_0$.



Callen Fig 5.2

$$E_0 = \min_X [E(S_0, X)]$$

We thus have two contrasting formulations:

entropy formulation: fundamental function is $S(E, X_1, X_2, \dots)$
if constraint on some X_i is removed, X_i will take the value that maximizes S for the fixed total energy E . In equilib, $d^2S < 0$. S ^{concave} ~~convex~~

energy formulation: fundamental function is $E(S, X_1, X_2, \dots)$
if constraint on some X_i is removed, X_i will take the value that minimizes E for the fixed total entropy S . In equilib $d^2E > 0$
 E ~~concave~~ ^{convex}

Suppose we had some equilib state for which E was not the minimum possible value for the given S .

Then can withdraw energy from the system by ^① doing mechanical work (for example drive a piston) while keeping S constant.

start E_0, S_0 $\xrightarrow[\text{work}]{\text{do mechanical (1) and heat (2)}}$ E_1, S_1
where $E_1 < E_0$ where $S_1 > S_0$

② now return this energy to the system in the form of heat $E_0 - E_1 = dQ = TdS$. The energy is now back to E_0 , but the entropy has increased by $dS = (E_0 - E_1)/T$.

The system is restored to its original energy but with a higher value of entropy. But this contradicts the requirement that the original equilibrium state was a maximum of entropy. \Rightarrow original E had to have been the minimum.

We have now two equivalent representations

- 1) entropy $S(E, V, N)$ energy E , volume V , number N
held fixed
- 2) energy $E(S, V, N)$ entropy S , volume V , number N
held fixed

In certain cases it is more natural to regard temperature T as held constant, rather than S ; or to regard pressure p as held constant, rather than V ; or to regard chemical potential μ as held constant, rather than N .

We therefore wish to develop new formulations of thermodynamics that will allow us to regard T , p , or μ as a fundamental variable rather than S , V , or N . These new formulations will lead to the Helmholtz and Gibbs free energies that play the role of ~~entropy~~ analogues to energy as the fundamental thermodynamic function of these new formulations.

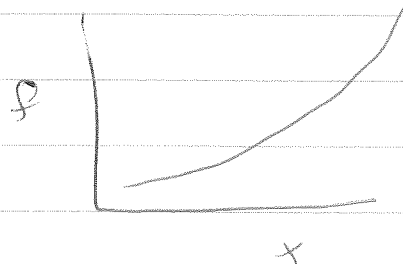
For example, we have $E(S, V, N)$ with $T = \left(\frac{\partial E}{\partial S}\right)_{V, N}$

How can we make a thermodynamic "potential" that contains all the information of $E(S, V, N)$ but depends on T rather than S .

Legendre Transformations

We treat this problem in general.

a general function $f(x)$



define the variable $p = \frac{df}{dx}$

How do we find a function that contains all the information in $f(x)$, but depends on p rather than x ?

First guess is just to invert $p(x) \equiv \frac{df}{dx}$ to solve for x as a function of p , i.e. $x(p)$. Then one could substitute this into $f(x)$ to get

$$g(p) = f(x(p))$$

This does not have the complete information contained in $f(x)$!

For example: $f = ax^2 + bx + c$

$$p = \frac{df}{dx} = 2ax + b \Rightarrow x = \frac{p-b}{2a}$$

$$g(p) = f(x(p)) = a\left(\frac{p-b}{2a}\right)^2 + b\left(\frac{p-b}{2a}\right) + c$$

$$= \frac{a}{4a^2} (p^2 - 2pb + b^2) + \frac{bp}{2a} - \frac{b^2}{2a} + c$$

$$= \frac{p^2}{4a} - \frac{b}{2a}p + \frac{b^2}{4a} + \frac{bp}{2a} - \frac{b^2}{2a} + c$$

$$g(p) = \frac{p^2}{4a} - \frac{b^2}{4a} + c$$

Consider now $f'(x) = a(x-x_0)^2 + b(x-x_0) + c$

$$= ax^2 - 2axx_0 + ax_0^2 + bx - bx_0 + c$$

$$= ax^2 + b'x + c'$$

where $b' = b - 2ax_0$

$c' = c - bx_0 + ax_0^2$

$\Rightarrow g'(p) = \frac{p^2}{4a} - \frac{b'^2}{4a} + c'$

$= \frac{p^2}{4a} - \frac{(b^2 - 4abx_0 + 4a^2x_0^2)}{4a} + c - bx_0 + ax_0^2$

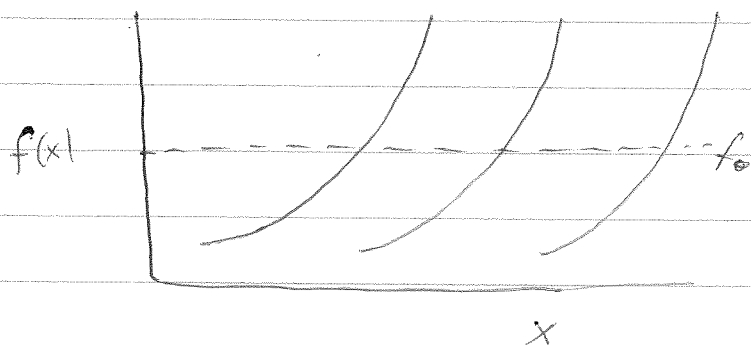
$= \frac{p^2}{4a} - \frac{b^2}{4a} + bx_0 - ax_0^2 + c - bx_0 + ax_0^2$

$= \frac{p^2}{4a} - \frac{b^2}{4a} + c$

$g'(p) = g(p)$

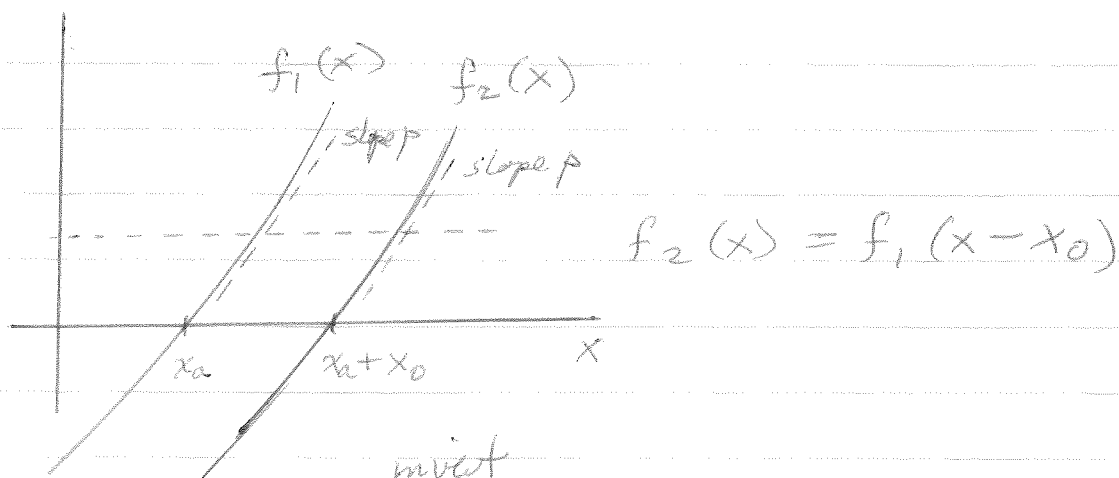
clearly $g(p)$ has lost some information since we get the same $g(p)$ for $f(x)$ and $f(x-x_0)$.

In general this is true: the procedure above cannot distinguish between $f(x)$ and $f(x-x_0)$ for any function $f(x)$.



← set of functions displaced from each other by fixed amount along x axis. For each function, the slope at constant $f = f_0$ is the same

Consider $f_1(x)$ and $f_2(x)$ defined as follows,



$$\text{Define } p_1(x) = \frac{df_1(x)}{dx} \xrightarrow{\text{invert}} x_1(p)$$

$$p_2(x) = \frac{df_2(x)}{dx} \xrightarrow{\text{invert}} x_2(p)$$

$$\frac{d^2 f_1(x - x_0)}{dx^2} = p_1(x - x_0)$$

$$\Rightarrow x_2(p) = x_1(p) + x_0$$

i.e. If $f_1(x)$ has slope p at x_1 , then $f_2(x)$ has slope p at $x_2 = x_1 + x_0$

$$g_1(p) = f_1(x_1(p))$$

$$g_2(p) = f_2(x_2(p)) = f_1(x_2(p) - x_0)$$

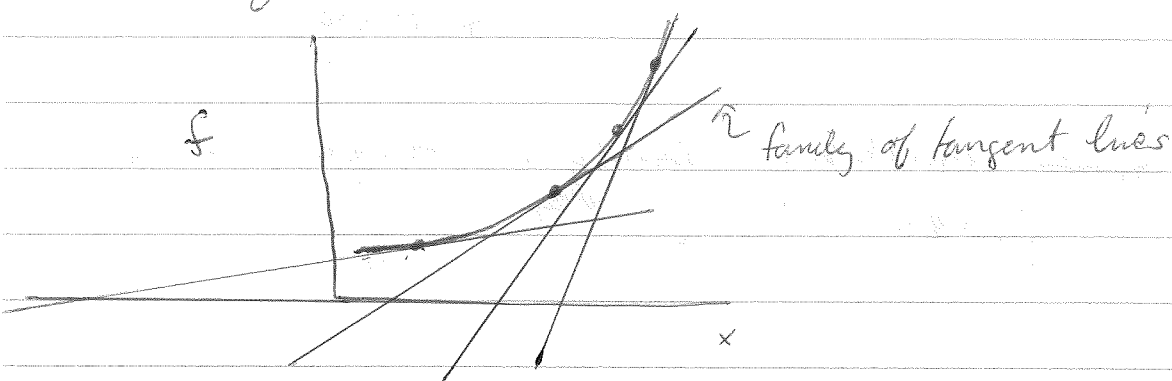
$$= f_1(x_1(p) + x_0 - x_0) = f_1(x_1(p))$$

$$= g_1(p)$$

So given $g(p) = g_1 = g_2$, we cannot tell if the original function was f_1 or f_2 ! So $g(p)$ does NOT have all the same information as the function $f(x)$.

hence writing the function as a function of the derivative $p = \frac{df}{dx}$, rather than x , results in the same $g(p)$ in each case.

However an alternate, correct, approach is given by noting that any curve can be described by the envelope of its tangent lines.



the line tangent to the curve $f(x)$ at point x_0 is given by the equation

$$y = px + b \quad \text{where} \quad p = \left. \frac{df}{dx} \right|_{x=x_0}$$

$$\text{and} \quad f(x_0) = px_0 + b \Rightarrow b = f(x_0) - px_0$$

b is the y -intercept, i.e. $y = b$ when $x = 0$.

Define the function

$$g(p) = f(x) - px$$

$$\text{where} \quad p = \frac{df}{dx}$$

gives the y -intercept of the tangent to the curve at the point where the curve has slope p

In above one solves $p(x) = \frac{df}{dx}$ to get the inverse function $x(p)$, and substitutes this $x(p)$ in above expression for g to get a

function of only p .

Alternatively, one can define $g(p)$ by

$$g(p) = \underset{x}{\text{extremum}} [f(x) - px]$$

↑ take the value of x that gives an extremum of $[f(x) - px]$

In this way, $g(p)$ is ^{clearly} independent of x , and the extremum condition guarantees that

$$\frac{df}{dx} - p = 0 \Rightarrow p = \frac{df}{dx}$$

When $f(x)$ is convex, i.e. $\frac{d^2f}{dx^2} > 0$, then the extremum is the minimum of $f - px$.

When $f(x)$ is concave i.e. $\frac{d^2f}{dx^2} < 0$, then the extremum is the maximum of $f - px$.

Note:

$$\frac{dg}{dp} = \frac{d}{dp} [f(x) - px] = \frac{df}{dx} \frac{dx}{dp} - x - p \frac{dx}{dp}$$

$$= \left[\frac{df}{dx} - p \right] \frac{dx}{dp} - x = 0 - x$$

$$= -x$$

$$\text{since } \frac{df}{dx} = p$$

To summarize

$$f(x) \quad \phi \equiv \frac{df}{dx}$$

$$g(p) = f(x) - px \quad \Rightarrow \quad \frac{dg}{dp} = -x$$

One says that $g(p)$ is the Legendre transform of $f(x)$ and that x and p are conjugate variables.

$g(p)$ contains all the information that $f(x)$ does.

ie. if one knows $g(p)$ then one can construct $f(x)$ from it, by constructing all the tangent lines $y = px + g(p)$.

The Legendre transform allows one to switch variables from x to $\frac{df}{dx}$ without losing information.

You may have already seen Legendre transforms in classical mechanics. In the Lagrange formulation, the fundamental function is the Lagrangian $\mathcal{L}[q, \dot{q}]$ which depends on the variables q and \dot{q} . In the Hamilton formulation one wants to replace the variable \dot{q} by the variable $\phi = \frac{\partial \mathcal{L}}{\partial \dot{q}}$. The fundamental function to use,

which is a function of q and ϕ rather than q and \dot{q} , is therefore the Legendre transform of the Lagrangian

$$\mathcal{L}[q, \dot{q}] - \phi \dot{q} = -\mathcal{H}[p, q]$$

where \mathcal{H} is the Hamiltonian. Because ϕ and q are conjugate variables, we know that