

# Lecture #13

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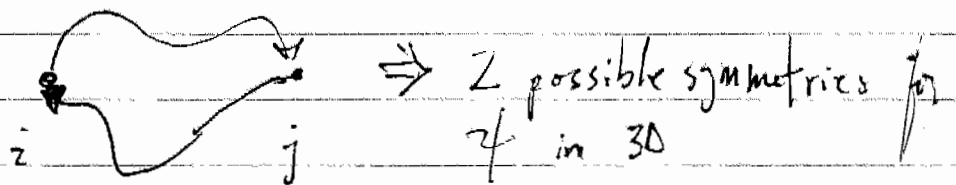
## Quantum Many Particle Systems:

$N$  identical particles described by a wavefunction

$$\Psi(\vec{r}_1, s_1; \vec{r}_2, s_2; \dots; \vec{r}_N, s_N) \quad \begin{array}{l} \vec{r}_i = \text{position of particle } i \\ s_i = \text{spin of } i \end{array}$$

Identical particles  $\Rightarrow$  prob. dist.  $|\Psi|^2$  should not change under exchange of any pair of coordinates:

$$|\Psi(\vec{r}_1, s_1; \dots; \vec{r}_i, s_i; j; \dots; \vec{r}_j, s_j; \dots; \vec{r}_N, s_N)|^2 = |\Psi(1, \dots, j; i; \dots; N, s_N)|^2$$



- 1)  $\Psi$  is symmetric under pair exchange.
- 2)  $\Psi$  is anti-symmetric

$$\Psi_S(1, 2, \dots, i, j, \dots, N) = +\Psi_S(1, \dots, j, \dots, i, \dots, N)$$

$$\Psi_A(1, \dots, i, j, \dots, N) = -\Psi_A(1, \dots, j, \dots, i, \dots, N)$$

- (1) particles called Bosons  $\rightarrow$  Bose-Einstein stats.
- (2) " " Fermions  $\rightarrow$  Fermi-Dirac "

Consider a general permutation  $\mathbb{P}$  (2)  
that interchanges any number of pairs of particles

$$(1) \text{ BE} \Rightarrow \mathbb{P}\psi = \psi$$

$$(2) \text{ FD} \Rightarrow \mathbb{P}\psi = (-1)^{\mathbb{P}}\psi, \text{ where}$$

$\mathbb{P} = \#$  of pair exchanges

$$\cancel{(-1)^{\mathbb{P}}} \quad (-1)^{\mathbb{P}} = \begin{cases} + & \text{for even permutation} \\ - & \text{for odd permutation} \end{cases}$$

Field Theory gives the spin/statistics theorem:

BE statistics  $\longleftrightarrow$  integer spin,  $s = 0, 1, 2, \dots$

FD statistics  $\longleftrightarrow$  half-integer spin,  $s = \frac{1}{2}, \frac{3}{2}, \dots$

Consider non-interacting particles.

$$H(1, 2, \dots, N) = H^{(1)}(1) + H^{(1)}(2) + H^{(1)}(3) + \dots$$

single particle Hamiltonians

$$\psi(1, 2, \dots, N) = \phi_1(1) \phi_2(2) \dots \phi_N(N), \quad \phi_i \text{ is an eigenstate}$$

of single-particle Hamiltonian  $H^{(1)}(i)$  w/ energy  $\epsilon_i$

Does not have proper symmetry.

$$BE: \psi = \frac{1}{\sqrt{N!}} \sum_{\{P\}} P \psi$$

↑ all possible # of permutations of N particles =  $\frac{N!}{1!}$

$$FD: \psi = \frac{1}{\sqrt{N!}} \sum_{\{P\}} (-1)^P P \psi$$

This gives the desired symmetry.

In Fermi case, it can be written as a Slater determinant of the single-particle states:

$$\psi_{FD} = \frac{1}{\sqrt{N!}} \begin{vmatrix} \phi_1(1) & \phi_1(2) & \dots & \phi_1(N) \\ \phi_2(1) & \phi_2(2) & \dots & \phi_2(N) \\ \vdots & \vdots & & \vdots \\ \phi_n(1) & \phi_n(2) & \dots & \phi_n(N) \end{vmatrix}$$

$$E = \sum_j n_j \epsilon_j ; n_j = \# \text{ of particles in state } j$$

For FD,  $n_j \in \{0, 1\}$  only. Why?

$$\psi(1, 2, \dots, N) = \phi_1(1) \phi_1(2) \phi_2(3) \dots \phi_n(N)$$

↑ ↑ particles 1 and 2 in same state  $\phi_1$

When we antisymmetrize:

$$\text{get } \phi_1(1) \phi_2(2) \dots + (-1) \phi_1(2) \phi_1(1) \dots = 0$$

for every term

# Pauli Exclusion Principle:

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No 2 Fermions can occupy the same state, or have the same quantum numbers

For BE stats  $\rightarrow$  no such restriction -  $n_j \in \mathbb{Z}$

Specification of non-interacting  $N$  particle quantum state is given by occupation numbers  $\{n_i\}$ . specifies 1 state of  $N$  particle system

Example: Hetero vs. Homo-nuclear Molecules:

Internal Modes of a Gas:  $E_j = E_{\text{trans}} + E_{\text{vib}} + E_{\text{rot}} + E_{\text{electronic}} + E_{\text{nuclear}}$

Non-interacting  $\Rightarrow$

$$Z = Z_{\text{trans}} Z_{\text{vib}} Z_{\text{rot}} Z_{\text{elec}} Z_{\text{nuclear}}$$

Focus on the rotational energy  $H = \frac{\hat{L}^2}{2I} \rightarrow E_l = \frac{l(l+1)\hbar^2}{2I}$

defn  $E_{\text{rot}} = l(l+1)\hbar^2 \Theta_{\text{rot}}^{-1}$ ;  $\Theta_{\text{rot}} \sim \frac{1}{k} = \frac{\hbar^2}{2Ik}$

Consider Heteronuclear atoms: (CO, HD, NO, ...)

with each energy eigenstate, there is the  $m_l$  degeneracy =  $2l+1$ , ( $m_l = (-l, -l+1, \dots, 0, \dots, l)$ )

$r_e = \text{equilibrium distance}$

$$I = \mu r_e^2$$

$$\mu = \frac{m_1 m_2}{m_1 + m_2}$$

$$V = \frac{\partial \ln Z}{\partial \beta}$$

$$C_V = \frac{\partial U}{\partial T}$$

degeneracy

$$Z_{rot} = \sum_{l=0}^{\infty} (2l+1) e^{-l(l+1) \frac{\theta_{rot}}{T}}$$

$$= \sum_{l=0}^{\infty} (2l+1) e^{-l(l+1) \frac{\theta_{rot}}{T}}$$

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$$\approx \int_0^{\infty} (2l+1) e^{-l(l+1) \frac{\theta_{rot}}{T}} dl$$

$T \gg \theta_{rot} \quad \frac{\theta_{rot}}{T} \ll 1$

$$Z_{rot} = (Z_{rot})^N \text{ for } N \text{ molecules.} \Rightarrow C_V = Nk = \frac{20}{5} R$$

Homonuclear Molecules ( $H_2, O_2, D_2, \dots$ )  
are more complicated (statistics / indistinguishability)

$$\Psi = \Psi_{electronic} \Psi_{vib} \Psi_{rot} \Psi_{nuclear}$$

$$P \Psi_{vib} = \Psi_{vib} \text{ (a function of } |R_1 - R_2| \text{)}$$

$$P \Psi_{electronic} = \pm \Psi_{electronic} \text{ (depending on the particular orbital)}$$

We classify orbitals by  $L$  component along the molecular axis.

Denote states by  $\Lambda = |M_L|$



So too with spin

$\Lambda$	name
0	$\Sigma$
1	$\Pi$
2	$\Delta$

think about symmetry with respect to particle exchange.

Recall the orbital wave functions are  $\Psi_{orbit} \propto Y_{l,m}(\theta, \phi)$  (rotational eigenstates)

$$Y_{l,m}(\theta, \phi) \propto P_l^m(\cos \theta) e^{im\phi}$$

Exchange of nuclei corresponds to  $\theta \rightarrow \pi - \theta$   
 $\phi \rightarrow \phi + \pi$

$$\therefore P_x^m [\cos(\pi - \theta)] = (-1)^{l+m} P_x^m [\cos \theta]$$

and  $e^{i(\phi + \pi)m} = (-1)^m e^{i\phi m}$

$$\Rightarrow \mathbb{P} \psi_{\text{rotation}} = (-1)^{l/2} \psi_{\text{rotation}}$$

Consider now  $H_2$ , and consider the nuclear spins.

Nucleus of 4 protons  $\rightarrow$  nuclear spin of  $I_1 = \frac{1}{2}$ ,

eigenstates of total nuclear spin  $I = I_1 + I_2$ ,  
correspond to  $I = 1, I_z = 1, 0, -1$  AND

$$I = 0, I_z = 0 \quad (\text{from QM}) ; |\psi\rangle = |I_1, I_{1z}\rangle |I_2, I_{2z}\rangle$$

symmetric (even)

$$\left\{ \begin{aligned} |1, 1\rangle &= |\frac{1}{2}, \frac{1}{2}\rangle_1 |\frac{1}{2}, \frac{1}{2}\rangle_2 \\ |1, 0\rangle &= \frac{1}{\sqrt{2}} [|\frac{1}{2}, \frac{1}{2}\rangle_1 |\frac{1}{2}, -\frac{1}{2}\rangle_2 + |\frac{1}{2}, -\frac{1}{2}\rangle_1 |\frac{1}{2}, \frac{1}{2}\rangle_2] \\ |1, -1\rangle &= |\frac{1}{2}, -\frac{1}{2}\rangle_1 |\frac{1}{2}, -\frac{1}{2}\rangle_2 \end{aligned} \right.$$

Antisymmetric (odd)

$$\left\{ \begin{aligned} |0, 0\rangle &= \frac{1}{\sqrt{2}} [|\frac{1}{2}, \frac{1}{2}\rangle_1 |\frac{1}{2}, -\frac{1}{2}\rangle_2 - |\frac{1}{2}, -\frac{1}{2}\rangle_1 |\frac{1}{2}, \frac{1}{2}\rangle_2] \end{aligned} \right.$$

have this definite symmetry because Hamiltonian commutes with  $\mathbb{P}$ .

The transition rates between nuclear spin states is small, so the two spin orientations are often considered as two different components (2 species of a gas) of  $H_2$  (7)

We specify the nuclear and rotational states as

$$\psi_{\text{tot}} = \psi_{\text{N.S.}}(I, I_z) \otimes \psi_{\text{rot}}(l, m) \otimes \psi_{\text{rest}} \leftarrow \text{symmetric under exchange.}$$

$$\therefore \Pi \psi_{\text{N.S.}} = \begin{cases} +1, & I=1 \\ -1, & I=0 \end{cases} \otimes \psi_{\text{N.S.}}$$

$$\Pi \psi_{\text{rot}} = (-1)^l \psi_{\text{rot}}(l, m)$$

$\Rightarrow$  if  $I=1$  and  $l=\text{even}$   
if  $I=0$  and  $l=\text{odd}$  } symmetric wavefunction  $\rightarrow$  forbidden because H is a Fermion.

$\Rightarrow$   $I=1$  and  $l=\text{odd}$   
 $I=0$  and  $l=\text{even}$  } anti-symmetric wavefunction  $\rightarrow$  allowed.

call para- $H_2$  the  $H_2$  w/ spin 0

ortho- $H_2$  the  $H_2$  w/ spin 1

$$\text{then } Z_{\text{para}} = \sum_{l=\text{even}}^{\infty} (2l+1) e^{-l(l+1)\theta_r/T}$$

$$Z_{\text{ortho}} = (3) \sum_{l=\text{odd}}^{\infty} (2l+1) e^{-l(l+1)\theta_r/T}$$

3 states for  $I_z = 1, 0, -1$

$$\therefore Z_{\text{rot}}^{\text{tot}} = Z_{\text{para}} + Z_{\text{ortho}}$$

(8)

Probability that a molecule is in the ortho (or para) state [regardless of  $l$ ] is given by

$$\text{Prob}[\text{ortho}] = \frac{Z_{\text{ortho}}}{Z_{\text{ortho}} + Z_{\text{para}}}; \text{Prob}[\text{para}] = \frac{Z_{\text{para}}}{Z_{\text{ortho}} + Z_{\text{para}}}$$

$$\text{as } T \gg \Theta_r; \sum_e \approx \int dl$$

$$\Rightarrow \text{Prob}(\text{ortho}) = \frac{3}{4}, \text{Prob}(\text{para}) = \frac{1}{4}$$