

Physics 114 - January 19, 2010

Reminders

■ No workshops this week
Begin Mon., Jan. 25

■ A-labs begin wk Jan. 25 - 28

■ B-labs begin wk Feb. 1 - 4

■ Problem Set 1 is posted on web
Due this Thursday



Any questions about
Syllabus/course organization?

Last Time - Electrostatics

■ two types of charge

Proportionality constant...
Not to be confused w/ k ...
Boltzmann's constant

■ Coulombs Law

$$k = \frac{1}{4\pi\epsilon_0} = \text{CONSTANT}$$

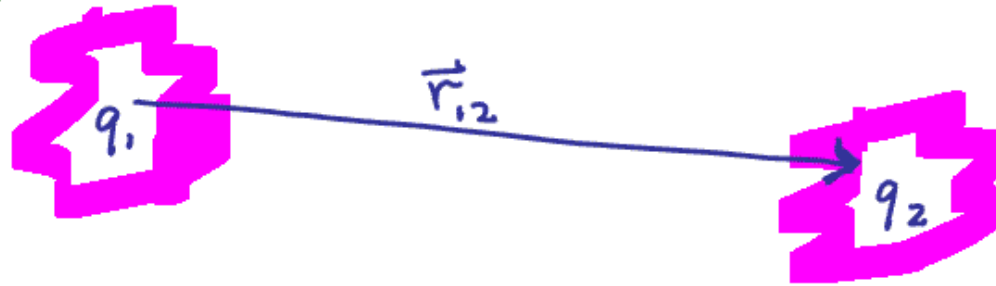
↑
Permittivity of Free space

$$\vec{F}_{12} = k \frac{q_1 q_2}{r_{12}^2} \hat{r}_{12}$$



Charles Coulomb

inverse square force... just like gravitation



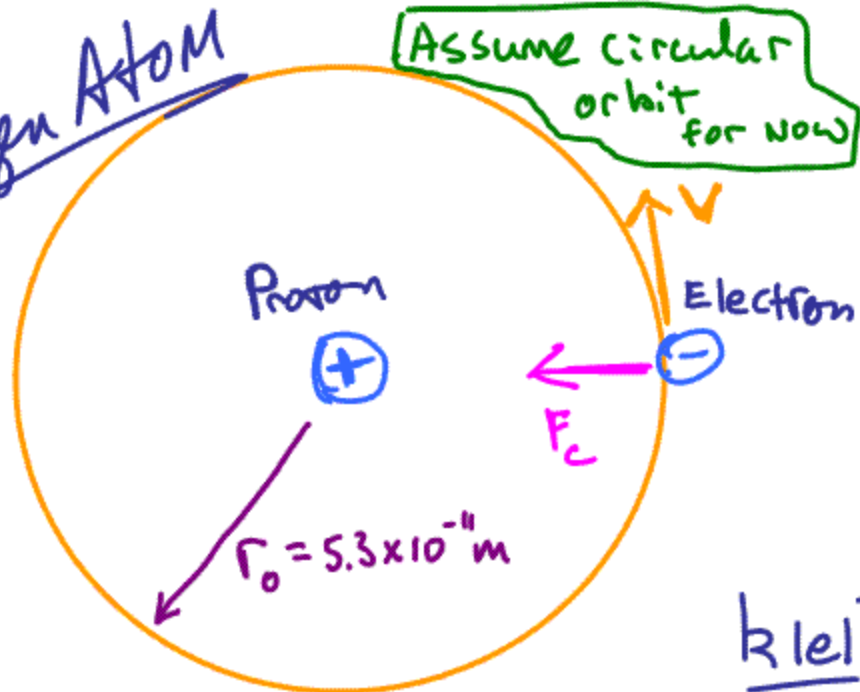
■ Electric charge is conserved

$$|e| = 1.6 \times 10^{-19} \text{ C}$$

■ Electric charge is quantized

$c \equiv \text{Coulomb}$
not to be conf. w/ speed of light c

Hydrogen Atom



F_c = Centripetal force

$$F_c = \frac{k |e| e^2}{r_0^2} \quad \text{Coulomb}$$

$$= \frac{m_e v^2}{r_0} \quad \text{Newton}$$

$$\frac{k |e| e^2}{r_0^2} = \frac{m_e v^2}{r_0}$$

Plugin $\left\{ \begin{array}{l} m_e = 9.11 \times 10^{-31} \text{ kg} \\ k = 8.99 \times 10^9 \frac{\text{Nm}^2}{\text{C}^2} \\ r_0 = 5.3 \times 10^{-11} \text{ m} \\ |e| = 1.6 \times 10^{-19} \text{ C} \end{array} \right\}$ Solve for v

Rough value for velocity of e^- in H atom in "Bohr model" ground state

later in term

$$v = 2.2 \times 10^6 \text{ m/s} \sim 1\% \text{ speed of light}$$

P e^- ionization E of hydrogen



$$\text{Energy} \sim \text{work} = \int \vec{F} \cdot d\vec{s}$$

$$\text{work} = \int_{r_0}^{\infty} \vec{F} \cdot d\vec{s} = \int_{r_0}^{\infty} \frac{k|e|^2}{r^2} dr = k|e|^2 \int_{r_0}^{\infty} \frac{dr}{r^2}$$

$$w = k|e|^2 \left[-\frac{1}{\infty} + \frac{1}{r_0} \right] = \frac{k|e|^2}{r_0} = 4.3 \times 10^{-18} \text{ J}$$

$$1 \text{ Joule} = 6.2 \times 10^{18} \text{ eV (electron-Volt)}$$

$$W = 26.9 \text{ eV}$$

True value
ionization E hydrogen
13.6 eV

e^- already moving - must take this into account

$$KE = \frac{1}{2} M_e v^2 = \frac{1}{2} (9.11 \times 10^{-31} \text{ kg}) (2.2 \times 10^6 \text{ m/s})^2$$
$$= 13.6 \text{ eV}$$

$$E_{\text{ionization}} = 26.9 \text{ eV} - 13.6 \text{ eV}$$
$$= 13.3 \text{ eV}$$

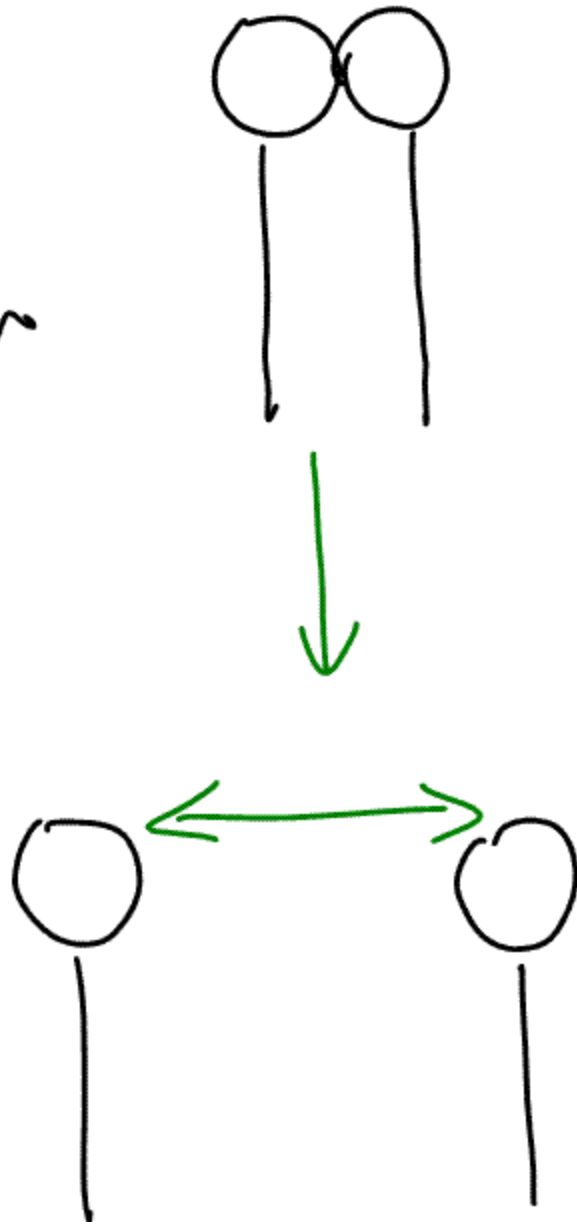
Fairly close to
Accepted
value



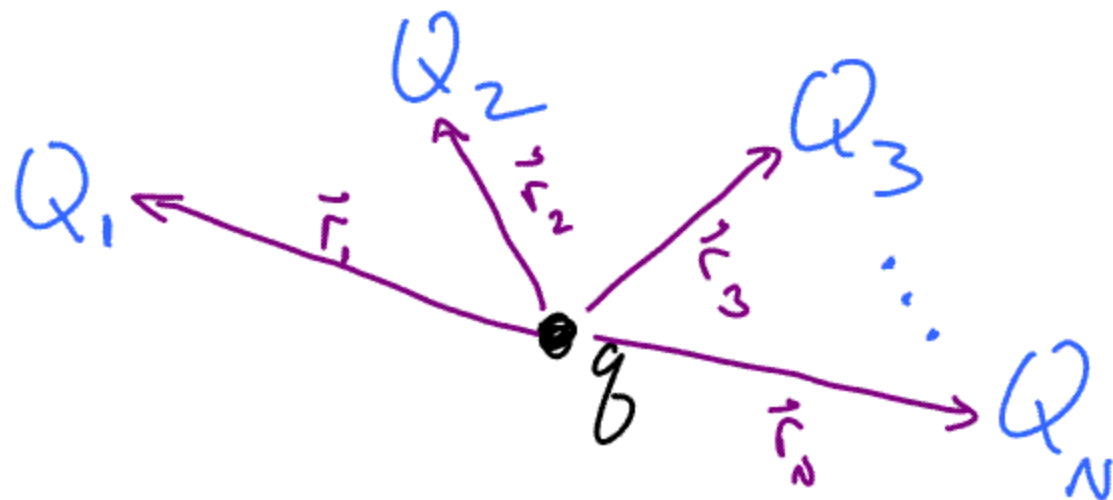
Wow!

Physics is sooo cool!

charging
by
induction
demo



Electrostatics is a Vector Force

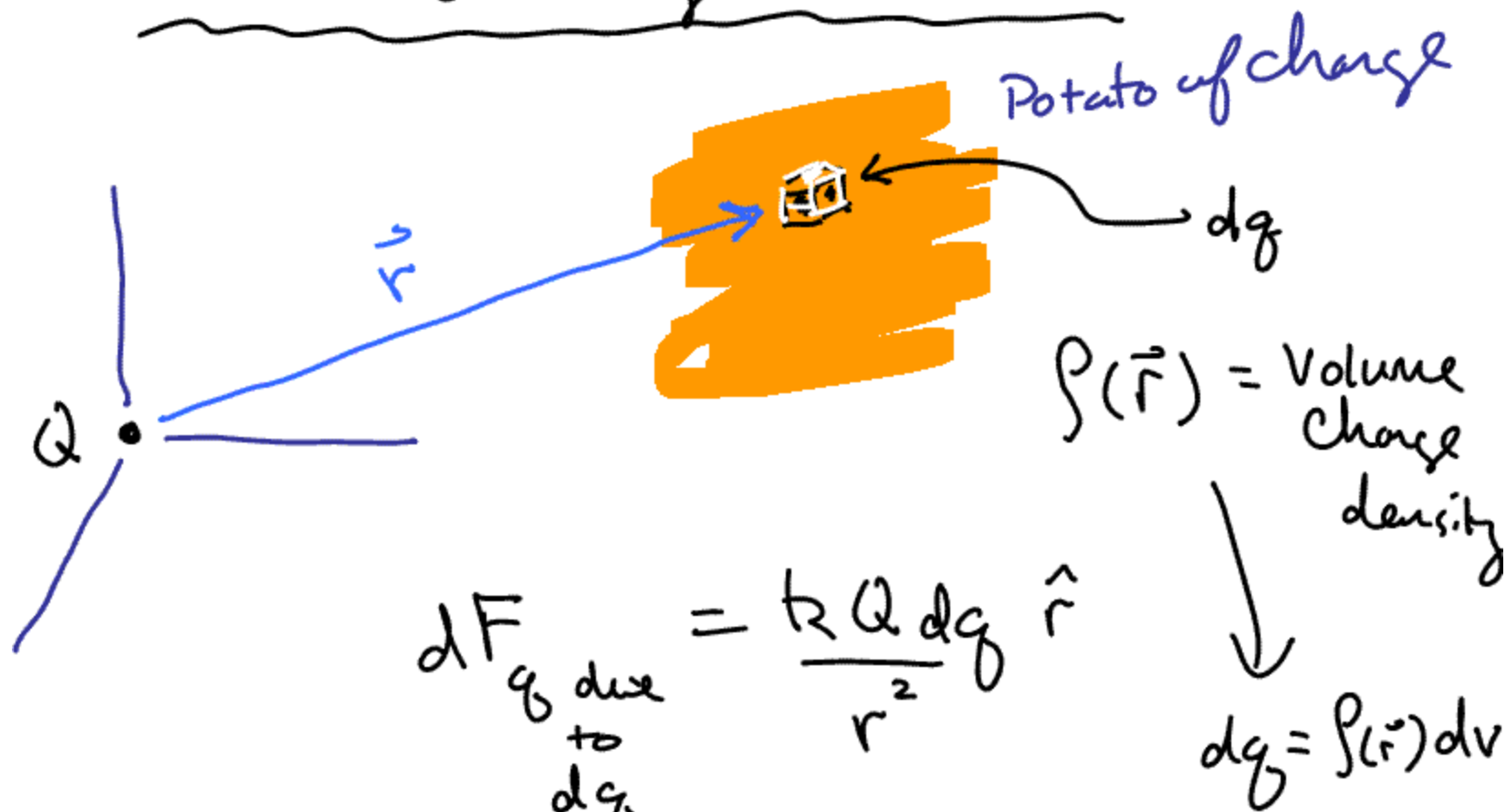


Force on q
due to
discrete
Point
charges

$$\vec{F}_{\text{net}} = \vec{F}_{Q_1} + \vec{F}_{Q_2} + \dots + \vec{F}_{Q_N}$$

$$\vec{F}_{\text{net}} = \frac{kQ_1q}{r_1^2} \hat{r}_1 + \frac{kQ_2q}{r_2^2} \hat{r}_2 + \dots + \frac{kQ_Nq}{r_N^2} \hat{r}_N$$

Continuous Charge distribution



$$dF_{q \text{ due to } dq} = \frac{kQ dq}{r^2} \hat{r}$$

$$\vec{F} = \int \frac{kQ dq}{r^2} \hat{r}$$

charge volume

$$\vec{F} = \int \frac{k Q \rho(\vec{r})}{r^2} dv \hat{r}$$

change
volume

Volume charge
distribution

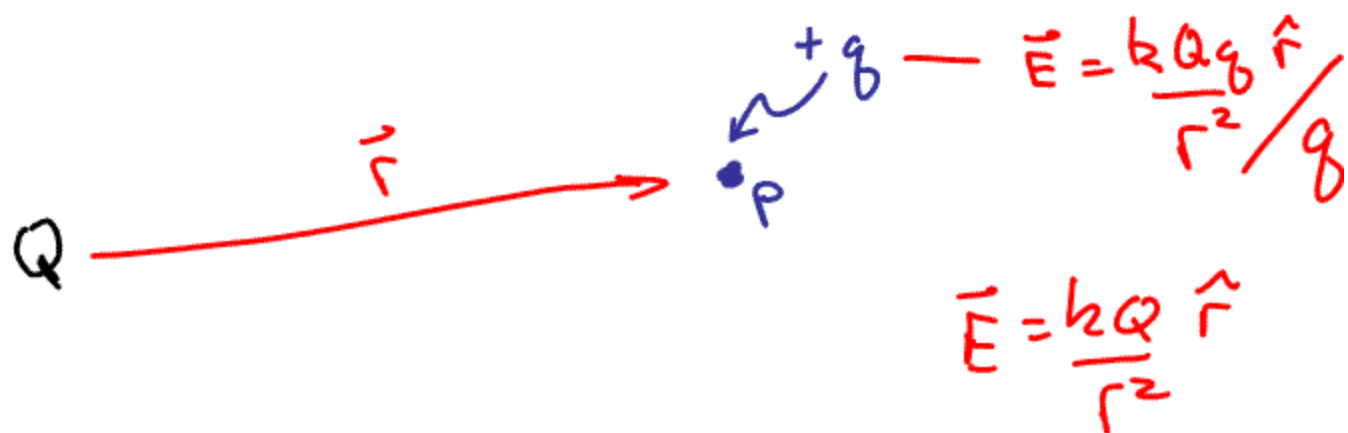
ρ vol charge density
Coul/m³

Plane or Surface
charge distribution

σ surf. chg density
Coul/m²

Linear charge distribution

λ linear chg
density
Coul/m



$\vec{E} \equiv$ Electric field $\equiv \frac{\vec{F}}{q} \equiv \frac{\text{Force}}{\text{charge}}$ on a
 '+' Test charge
 placed at
 the given
 Point

Fields are nothing new to you. . .

