

Response functions

specific heat at const volume $C_V = \left(\frac{dQ}{dT}\right)_{V,N} = T \left(\frac{dS}{dT}\right)_{V,N}$

specific heat at const pressure $C_P = \left(\frac{dQ}{dT}\right)_{P,N} = T \left(\frac{dS}{dT}\right)_{P,N}$

isothermal compressibility $K_T = -\frac{1}{V} \left(\frac{\partial V}{\partial P}\right)_{T,N}$

adiabatic compressibility $K_S = -\frac{1}{V} \left(\frac{\partial V}{\partial P}\right)_{S,N}$

coefficient of thermal expansion $\alpha = \frac{1}{V} \left(\frac{\partial V}{\partial T}\right)_{P,N}$

All the above may be viewed as a second derivative of an appropriate thermodynamic potential

$$C_V = T \left(\frac{dS}{dT}\right)_{V,N} = -T \left(\frac{\partial^2 A}{\partial T^2}\right)_{V,N} \quad \text{since } \left(\frac{\partial A}{\partial T}\right)_{V,N} = -S(T, V, N)$$

$$C_P = T \left(\frac{dS}{dT}\right)_{P,N} = -T \left(\frac{\partial^2 G}{\partial T^2}\right)_{P,N} \quad \text{since } \left(\frac{\partial G}{\partial T}\right)_{P,N} = -S(T, P, N)$$

$$K_T = -\frac{1}{V} \left(\frac{\partial V}{\partial P}\right)_{T,N} = -\frac{1}{V} \left(\frac{\partial^2 G}{\partial P^2}\right)_{T,N} \quad \text{since } \left(\frac{\partial G}{\partial P}\right)_{T,N} = V(T, P, N)$$

$$K_S = -\frac{1}{V} \left(\frac{\partial V}{\partial P}\right)_{S,N} = -\frac{1}{V} \left(\frac{\partial^2 H}{\partial P^2}\right)_{S,N} \quad \text{since } \left(\frac{\partial H}{\partial P}\right)_{S,N} = V(S, P, N)$$

$$\alpha = \frac{1}{V} \left(\frac{\partial V}{\partial T}\right)_{P,N} = \frac{1}{V} \left(\frac{\partial^2 G}{\partial T \partial P}\right)_N \quad \text{since } \left(\frac{\partial G}{\partial P}\right)_{T,N} = V(T, P, N)$$

Since all the various thermodynamic potentials can all be derived from one another, the various second derivatives must ~~not~~ be related. If we consider

cases where N is held constant (as in all the above response functions) then there ~~are only~~ can be only three independent second derivatives, for example

$$\left(\frac{\partial^2 G}{\partial T^2}\right)_{P,N} = -C_P/T$$

$$\left(\frac{\partial^2 G}{\partial P^2}\right)_{T,N} = -V R T$$

$$\left(\frac{\partial^2 G}{\partial T \partial P}\right)_N = V \alpha$$

All the other second derivatives of the other potentials must be some combination of these three.

Consider C_V we will show how to write it in terms of the above.

Consider Helmholtz free energy $A(T, V)$ since N is kept constant, we will not write it

$$-S(T, V) = \left(\frac{\partial A}{\partial T}\right)_V$$

Viewing S as a function of T , and V we have

$$dS = \left(\frac{\partial S}{\partial T}\right)_V dT + \left(\frac{\partial S}{\partial V}\right)_T dV$$

$$\Rightarrow T \left(\frac{\partial S}{\partial T}\right)_P = T \left(\frac{\partial S}{\partial T}\right)_V + T \left(\frac{\partial S}{\partial V}\right)_T \left(\frac{\partial V}{\partial T}\right)_P$$

$$\Rightarrow C_p = C_V + T \left(\frac{\partial S}{\partial V} \right)_T \left(\frac{\partial V}{\partial T} \right)_p$$

$$\text{Now } \left(\frac{\partial S}{\partial V} \right)_T = - \frac{\partial^2 A}{\partial T \partial V} = \left(\frac{\partial P}{\partial T} \right)_V$$

and $\left(\frac{\partial P}{\partial T} \right)_V \left(\frac{\partial T}{\partial V} \right)_p \left(\frac{\partial V}{\partial P} \right)_T = -1 \quad \leftarrow \begin{array}{l} \text{(see general result)} \\ \text{next page.} \end{array}$

Note: P, T, V are not independent

$$P = \left(\frac{\partial A}{\partial V} \right)_T \stackrel{?}{=} p(T, V)$$

$$\text{So } \left(\frac{\partial P}{\partial T} \right)_V = \frac{-1}{\left(\frac{\partial T}{\partial V} \right)_p \left(\frac{\partial V}{\partial P} \right)_T} = - \frac{\left(\frac{\partial V}{\partial T} \right)_p}{\left(\frac{\partial V}{\partial P} \right)_T}$$

$$C_p = C_V - T \left(\frac{\partial V}{\partial T} \right)_p \frac{\left(\frac{\partial V}{\partial T} \right)_p}{\left(\frac{\partial V}{\partial P} \right)_T}$$

$$= C_V - T \frac{(V\alpha)^2}{-V K_T} = C_V + TV \frac{\alpha^2}{K_T}$$

$$\text{So} \boxed{C_V = C_p - \frac{TV\alpha^2}{K_T}}$$

A general result for partial derivatives

Consider any three variables satisfying a constraint

$f(x, y, z) = 0 \Rightarrow z$ for example, is function of x and y
or y is function of x, z etc.

\Rightarrow exists a relation between partial derivatives of the
variables with respect to each other.

$$\text{constraint} \Rightarrow df = \left(\frac{\partial f}{\partial x}\right)_{y,z} dx + \left(\frac{\partial f}{\partial y}\right)_{x,z} dy + \left(\frac{\partial f}{\partial z}\right)_{x,y} dz = 0$$

if hold z const, i.e. $dz = 0$, then

$$\left(\frac{\partial x}{\partial y}\right)_z = - \frac{(\partial f/\partial y)_{x,y}}{(\partial f/\partial x)_{y,z}}$$

if hold y const, i.e. $dy = 0$, then

$$\left(\frac{\partial y}{\partial x}\right)_y = - \frac{(\partial f/\partial x)_{y,z}}{(\partial f/\partial z)_{y,x}}$$

if hold x const, i.e. $dx = 0$, then

$$\left(\frac{\partial z}{\partial y}\right)_x = - \frac{(\partial f/\partial z)_{x,y}}{(\partial f/\partial y)_{x,z}}$$

Multiplying together we get

$$\left(\frac{\partial x}{\partial y}\right)_z \left(\frac{\partial y}{\partial z}\right)_x \left(\frac{\partial z}{\partial x}\right)_y = -1$$

(x, y, z) with constraint among them

Solve for $x(y, z)$ or for $y(x, z)$

then $dx = \left(\frac{\partial x}{\partial y}\right)_z dy + \left(\frac{\partial x}{\partial z}\right)_y dz$ ①

$$dy = \left(\frac{\partial y}{\partial x}\right)_z dx + \left(\frac{\partial y}{\partial z}\right)_x dz \quad ②$$

Suppose way dx keeping $dz = 0$

$$\textcircled{1} \Rightarrow dx = \left(\frac{\partial x}{\partial y}\right)_z dy \Rightarrow \frac{dy}{dx} = \frac{1}{\left(\frac{\partial x}{\partial y}\right)_z}$$

$$\textcircled{2} \Rightarrow dy = \left(\frac{\partial y}{\partial x}\right)_z dx \Rightarrow \frac{dx}{dy} = \left(\frac{\partial x}{\partial y}\right)_z$$

$$\Rightarrow \boxed{\left(\frac{\partial y}{\partial x}\right)_z = \frac{1}{\left(\frac{\partial x}{\partial y}\right)_z}}$$

Similarly we must be able to write k_s in terms of c_p, k_T, α

Consider enthalpy $H(s, p)$

$$\left(\frac{\partial H}{\partial P}\right)_S = V(s, p)$$

regarding V as a function of s and p we have

$$dV = \left(\frac{\partial V}{\partial P}\right)_S dP + \left(\frac{\partial V}{\partial S}\right)_P ds$$

$$-\frac{1}{V} \left(\frac{\partial V}{\partial P}\right)_T = -\frac{1}{V} \left(\frac{\partial V}{\partial P}\right)_S - \frac{1}{V} \left(\frac{\partial V}{\partial S}\right)_P \left(\frac{\partial S}{\partial P}\right)_T$$

$$k_T = k_s - \frac{1}{V} \left(\frac{\partial V}{\partial S}\right)_P \left(\frac{\partial S}{\partial P}\right)_T$$

$$\text{Now } \left(\frac{\partial S}{\partial P}\right)_T = -\frac{\partial^2 G}{\partial T \partial P} = -\left(\frac{\partial V}{\partial T}\right)_P$$

$$\text{and } \left(\frac{\partial V}{\partial S}\right)_P = \frac{\left(\frac{\partial V}{\partial T}\right)_P}{\left(\frac{\partial S}{\partial T}\right)_P}$$

$$\text{above follows from: } \frac{\partial G}{\partial P} = V(T, P) \Rightarrow dV = \left(\frac{\partial V}{\partial T}\right)_P dT + \left(\frac{\partial V}{\partial P}\right)_T dP$$

$$-\frac{\partial G}{\partial T} = S(T, P) \Rightarrow dS = \left(\frac{\partial S}{\partial T}\right)_P dT + \left(\frac{\partial S}{\partial P}\right)_T dP$$

$$\Rightarrow \left(\frac{\partial V}{\partial S}\right)_P = \frac{\left(\frac{\partial V}{\partial T}\right)_P}{\left(\frac{\partial S}{\partial T}\right)_P}$$

or in general, if z and y are functions of u and x , i.e. $\left\{ z(u, x), y(u, x) \right\}$ then $\left(\frac{\partial z}{\partial y}\right)_x = \frac{\left(\frac{\partial z}{\partial u}\right)_x}{\left(\frac{\partial y}{\partial u}\right)_x}$

substitute in to get

$$k_T = k_S + \frac{1}{V} \left(\frac{\partial V}{\partial T} \right)_P \frac{\left(\frac{\partial V}{\partial T} \right)_P}{\left(\frac{\partial S}{\partial T} \right)_P} = k_S + \frac{1}{V} \frac{(V\alpha)^2}{C_P/T}$$

$$k_T = k_S + \frac{TV\alpha^2}{C_P}$$

$$k_S = k_T - \frac{TV\alpha^2}{C_P}$$

See Callen for a systematic way to reduce all such derivatives to combinations of C_P , k_T , α

The main point is not to remember how to do this, but that it can be done! There are only a finite number of independent 2nd derivatives of the thermodynamic potentials! [if consider only those N fixed, there are only C_P , k_T , α]

Another useful relation

$$C_V = T \left(\frac{dS}{dT} \right)_V$$

$$\text{since } dE = TdS - pdV \quad (N \text{ fixed})$$

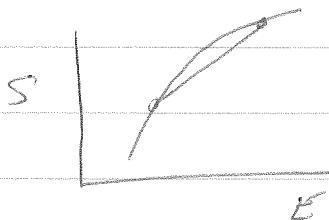
it follows that

$$C_V = \left(\frac{dE}{dT} \right)_V = T \left(\frac{dS}{dT} \right)_V$$

Stability

We already saw that the condition of stability required that $S(E)$ be a concave function

$$\frac{\partial^2 S}{\partial E^2} \leq 0.$$



concave \Leftrightarrow the chord drawn between any two points on curve lies below the curve

In a similar way, one can show $\frac{\partial^2 S}{\partial V^2} \leq 0$,

or more generally, S is concave in three dimensional S, E, V space

$$S(E + \Delta E, V + \Delta V, N) + S(E - \Delta E, V - \Delta V, N) \leq 2 S(E, V, N)$$

expanding the ~~left~~ ^{left} hand side in a Taylor series we get

$$\frac{\partial^2 S}{\partial E^2} \Delta E^2 + 2 \frac{\partial^2 S}{\partial E \partial V} \Delta E \Delta V + \frac{\partial^2 S}{\partial V^2} \Delta V^2 \leq 0$$

For $\Delta V = 0$ this gives $\frac{\partial^2 S}{\partial E^2} \leq 0$

For $\Delta E = 0$ this gives $\frac{\partial^2 S}{\partial V^2} \leq 0$

More generally, for ΔE and ΔV both $\neq 0$, we can rewrite as

$$(\Delta E, \Delta V) \begin{pmatrix} \frac{\partial^2 S}{\partial E^2} & \frac{\partial^2 S}{\partial E \partial V} \\ \frac{\partial^2 S}{\partial E \partial V} & \frac{\partial^2 S}{\partial V^2} \end{pmatrix} \begin{pmatrix} \Delta E \\ \Delta V \end{pmatrix} \leq 0$$

both eigenvalues of the matrix must be ≤ 0

That the quadratic form is always negative implies that
and so the determinant of the matrix ~~is negative~~ must be positive ≥ 0

$$\frac{\partial^2 S}{\partial E^2} \frac{\partial^2 S}{\partial V^2} - \left(\frac{\partial^2 S}{\partial E \partial V} \right)^2 \geq 0$$

Note: $\left(\frac{\partial^2 S}{\partial E^2} \right)_V = \frac{\partial}{\partial E} \left(\frac{1}{T} \right)_V = -\frac{1}{T^2} \left(\frac{\partial T}{\partial E} \right)_V = -\frac{1}{T^2 C_V}$

so $\left(\frac{\partial^2 S}{\partial E^2} \right)_V \leq 0 \Rightarrow C_V \geq 0$ specific heat is positive

Other Potentials

One can use the minimization principles of the other thermodynamic potentials, E, A, G , etc to derive other stability criteria.

Energy

S is maximum $\rightarrow E$ is minimum

S concave $\rightarrow E$ is convex

$$\Rightarrow E(S + \Delta S, V + \Delta V, N) + E(S - \Delta S, V - \Delta V, N) \geq 2E(S, V, N)$$

for $\Delta V = 0$

for $\Delta S = 0$

$$\Rightarrow \left(\frac{\partial^2 E}{\partial S^2} \right)_V = \left(\frac{\partial T}{\partial S} \right)_V \geq 0 \quad \text{and} \quad \left(\frac{\partial^2 E}{\partial V^2} \right)_S = -\left(\frac{\partial P}{\partial V} \right)_S \geq 0$$

and $\left(\frac{\partial^2 E}{\partial S^2} \right) \left(\frac{\partial^2 E}{\partial V^2} \right) - \left(\frac{\partial^2 E}{\partial S \partial V} \right)^2 \geq 0$

or $-\left(\frac{\partial T}{\partial S} \right)_V \left(\frac{\partial P}{\partial V} \right)_S - \left(\frac{\partial T}{\partial V} \right)_S^2 \geq 0$

using $\left(\frac{\partial T}{\partial S}\right)_V = \frac{T}{C_V} \rightarrow \left(\frac{\partial P}{\partial V}\right)_S = -\frac{1}{V K_S} \rightarrow \left(\frac{\partial V}{\partial P}\right)_S$

we get

$$\frac{T}{V C_V K_S} \geq \left(\frac{\partial T}{\partial V}\right)_S^2$$